

1 UNIVERSITY OF CALIFORNIA, SAN DIEGO

2 **Characterization and Modeling of Ion Transport Kinetics in p-Si Photovoltaic Modules**

3 A dissertation submitted in partial satisfaction of the  
4 requirements for the degree  
5 Doctor of Philosophy

6 in

7 NanoEngineering

8 by

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University of California, San Diego

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2020

## DEDICATION

25

To my family, who has supported me through this journey.

EPIGRAPH

26

27

*The world stands out on either side*

28

*No wider than the heart is wide;*

29

*Above the world is stretched the sky,—*

30

*No higher than the soul is high.*

31

—Edna St. Vincent Millay

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## 90 **List of Abbreviations**

91 **AC** Alternating current.

92 **Al-BSF** Aluminum back-surface field.

93 **ARC** Anti-reflective coating.

94 **BTS** Bias-temperature-stress.

95 **C-V** Capacitance-voltage.

96 **DC** Direct current.

97 **EDX** Energy dispersive X-ray spectroscopy.

98 **EVA** Ethylene-vinyl acetate.

99 **FEM** Finite element method.

100 **HRTEM** High-resolution transmission electron microscopy.

101 **J-V** Current-density-voltage.

102 **MIS** Metal-insulator-semiconductor.



- 103 **n-Si** *n*-type silicon.
- 104 **n-Si** *p*-type silicon.
- 105 **n<sup>+</sup>-Si** Highly-doped *n*-type silicon.
- 106 **PDE** Partial differential equation.
- 107 **PECVD** Plasma-enhanced chemical vapor deposition.
- 108 **PID** Potential-induced degradation.
- 109 **PID-s** Potential-induced degradation of the shunting type.
- 110 **PNP** Poisson-Nernst-Planck.
- 111 **PV** Photovoltaic.
- 112 **SF** Stacking Fault.
- 113 **SIMS** Secondary-ion mass spectroscopy.
- 114 **ToF-SIMS** Time-of-flight secondary-ion mass spectroscopy.
- 115 **XPS** X-ray photoelectron spectroscopy.

## 116 **List of Symbols**

- 117  $\epsilon_0$  Permittivity of free space
- 118  $\epsilon_r$  Relative permittivity
- 119  $\mu$  Mobility (ionic or electronic)
- 120  $\nabla$  The Nabla operator
- 121  $\phi$  Electric Potential
- 122  $\mathbf{E}$  Electric field
- 123  $C$  Concentration
- 124  $k_B$  Boltzmann's constant
- 125  $q$  Elementary charge
- 126  $Q_S$  Image charge in the semiconductor.
- 127  $R_{sh}$  Shunt resistance
- 128  $T$  Temperature
- 129  $V$  Voltage

130  $V_{FB}$  Flatband voltage

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264 ABSTRACT OF THE DISSERTATION

265 **Characterization and Modeling of Ion Transport Kinetics in p-Si Photovoltaic Modules**

266 by

267 Erick Rolando Martinez Loran

268 Doctor of Philosophy in NanoEngineering

269 University of California San Diego, 2020

270 Professor Prabhakar R. Bandaru, Chair

271 Though generally reliable, silicon solar modules can be subject to unforeseen degradation, leading  
272 to a duty life shorter than the expected 25-year life cycle. Potential-induced degradation (PID) has  
273 proven difficult to characterize and study. This dissertation is dedicated to developing a physical  
274 model to understand the kinetics of PID of the shunting type, and explain the factors that may lead  
275 to the design of PID-robust modules.

276 A bias-temperature stress (BTS) methodology to study ion migration in dielectric films is  
277 presented, which accounts for the contribution of bulk traps in the dielectric. Using this method,  
278 an Arrhenius relationship for the diffusivity of  $\text{Na}^+$  in  $\text{SiN}_x$  is determined, for which the prefactor  
279 is  $D_0 = 1.4 \times 10^{-14} \text{cm}^2/\text{s}$ , and the activation energy is  $E_a = 0.14 \text{eV}$ , with a 95 % confidence  
280 interval of [0.07, 0.21] eV. Based on this result, we bound the transit time of sodium ions, through  
281 highly resistive  $\text{SiN}_x$  anti-reflective coatings, within 1 h and 2 d, under temperature and electric  
282 fields relevant to PV operation.

283 A numerical solution to the coupled Poisson-Nernst-Planck system of equations is pre-  
284 sented, based on the finite element method (FEM), that can accurately simulate ionic transport in  
285 dielectrics and stacks of materials. The FEM implementation adequately describes the accumula-  
286 tion of charge in the semiconductor interface of metal-insulator-semiconductor capacitors (MIS).  
287 Using this model, we evaluate diffusion coefficients of  $\text{Na}^+$  in  $\text{SiO}_2$  under BTS conditions.

288 A methodology to simulate PID degradation in PV modules is derived, which uses the result  
289 from the ion transport model to simulate the characteristic J-V of the devices. PID is adequately  
290 described by the presence of metallic shunt at the  $p$ - $n$  junction of the cell, for which, the metal  
291 conductivity depends on the sodium concentration. An upper bound for the diffusivity of Na in  
292 stacking faults that result in PID is estimated to be  $10^{-14} \text{ cm}^2/\text{s}$ , based on comparison of the  
293 simulated PID time series with experimental reports of PID-s.

# 1 | Introduction

The photovoltaic industry has made outstanding progress over the last decades to bring the promise of a low-carbon economy closer than ever. One of the main advantages of photovoltaic energy is the low cost of operation inherent to silicon photovoltaic (PV) modules, which largely compensates for the considerable initial investment of solar developments. Silicon PV modules are required to operate with minimal power degradation over an average lifespan of 25 years in order to meet commercial warranties [1, 2]. Driven by this concern, reliability studies have become a subject of interest. Several procedures have been developed to test the reliability of modules under normal operation and under accelerated conditions. The latter are devised to forecast the performance of PV modules over the course of many years; provided that the acceleration protocols adequately relate to long term PV module operation. While some of the mechanisms of degradation seem evident (e.g. mechanical failure, loss of transparency of the encapsulant, delamination, shattering, among others) and have been addressed, the type of degradation studied in this dissertation is intrinsically more complex and remediation requires comprehensive knowledge of the properties of all the materials in the module.

A brief introduction on potential-induced degradation of the shunting type is given in this chapter, followed by an overview, in chapter 2, of the transport of Na in PV modules.

311 In chapter 3, a quantification methodology for the transport of Na in silicon nitride ( $\text{SiN}_x$ )  
312 is be presented. It is be shown that, for low ionic concentrations with respect to the applied elec-  
313 tric stress, the kinetics of Na transport can be described as a diffusion-advection problem. A  
314 methodology to quantify ion migration in  $\text{SiN}_x$  metal-insulator-semiconductor (MIS) capacitors  
315 is be demonstrated. A detailed analysis on the characteristic transit time is be presented, beyond  
316 which accumulation at the semiconductor interface occurs and leakage of Na to the silicon can  
317 occur at a larger scale.

318 In chapter 4, a finite element solution to the coupled Poisson-Nernst-Planck system of  
319 equations is derived, which is used to model transport of charged species in metal-insulator-  
320 semiconductor devices and other stacks of materials, beyond the assumptions of low contaminant  
321 concentration with respect to the applied electrical stress. A comparison with experimental data is  
322 given which agrees well with the predictions of the model.

323 In chapter 5, a modeling framework is presented, which mechanistically describes the ki-  
324 netics of PID-s degradation in p-Si modules, based on the bill of materials in the PV module and  
325 the operating conditions. Based on the results of the model, and comparison with PID reports, an  
326 upper bound for the diffusion coefficient of Na in the stacking fault is be estimated.

## 327 **1.1 What is Potential-Induced Degradation?**

328 Solar panels incorporate several individual solar cells connected in series, in order to  
329 achieve typical load requirements of several hundreds to thousands of volts. Fig. 1.1, shows a  
330 simplified schematic of a solar module, where each individual cell is connected in series to achieve  
331 a load of 1000 V. As these devices are mounted on a metal frame that is grounded, the voltage

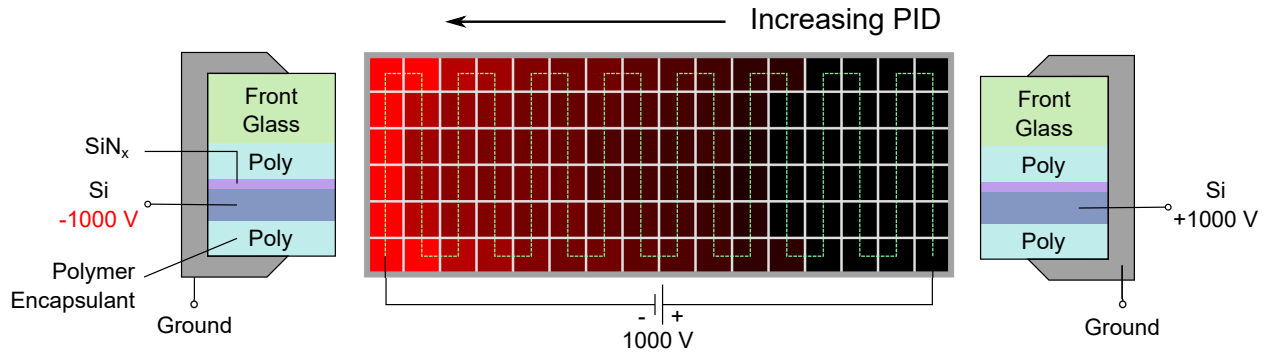


Figure 1.1: Schematic of a PID degraded silicon solar module. Reliability tests reveal that several cells close to the frame on the side of the string with negative floating polarity, are subject PID failure.

332 differential between the metal frame ( $V = 0$ ) and the floating potential of the cells becomes con-  
 333 siderable at the extremes of the array. Also in Fig. 1.1, a simplified schematic of the cross-section  
 334 of a mono-facial module is shown, consisting of a solar cell with a  $\text{SiN}_x$  anti-reflective coating  
 335 (ARC), encapsulated within a polymer (typically ethylene-vinyl acetate or EVA), and a glass win-  
 336 dow on the front side, usually made from soda lime glass. It has been observed that the cells  
 337 subject to a more negative potential with respect to the metal frame suffer from degradation in the  
 338 output power [3–5]. This degradation mode is called potential-induced degradation (PID) and is  
 339 known to affect  $n$ -type silicon ( $n$ -Si) and  $p$ -type silicon ( $n$ -Si) devices differently. In the  $n$ -Si case,  
 340 it has been reported that degradation is the result of surface polarization and is almost completely  
 341 reversible [6], whereas in  $p$ -Si, the degradation is due to a reduction in the shunt resistance  $R_{sh}$  of  
 342 the device [5] as is illustrated in Fig 1.2. While recovery has been documented for  $p$ -Si modules  
 343 affected by PID [5, 7–15], in general, only partial recovery is observed, and it requires the appli-  
 344 cation of a temperature stress and/or inverting the polarity of the voltage between the cell and the  
 345 frame. This makes the study of this type degradation on  $p$ -Si of special interest, and is the focus of  
 346 this research.

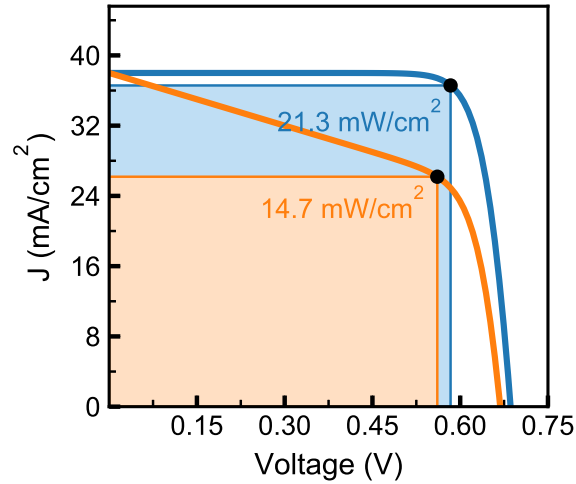


Figure 1.2:  $J$ - $V$  signature of a p-Si module affected by PID-s.

347 PID in p-Si occurs as a result of several factors facilitating the formation of electrical shunts  
 348 in the emitter of the device. Initial investigations suggested corrosion played an important role in  
 349 degradation [3], which led to the characterization of leakage currents across the module stack,  
 350 aimed at determining the physical pathways of the degradation process [5, 6, 16, 17]. These  
 351 experiments helped to identify ionic currents and, particularly, sodium ions ( $\text{Na}^+$ ) as the main  
 352 factor involved in PID in p-Si solar modules [17, 18].

## 353 1.2 Root cause of PID

354 The mechanism by which Na induces shunting has been elucidated by a combination of en-  
 355 ergy dispersive X-ray spectroscopy (EDX) and high-resolution transmission electron microscopy  
 356 (HRTEM) studies performed at the Fraunhofer Center for Silicon Photovoltaics, which showed  
 357 incorporation of Na inside stacking faults (SFs) in the silicon emitter, on severely shunted regions  
 358 of PID-affected devices [9, 12, 13, 19]. These findings indicate that Na introduces a large con-

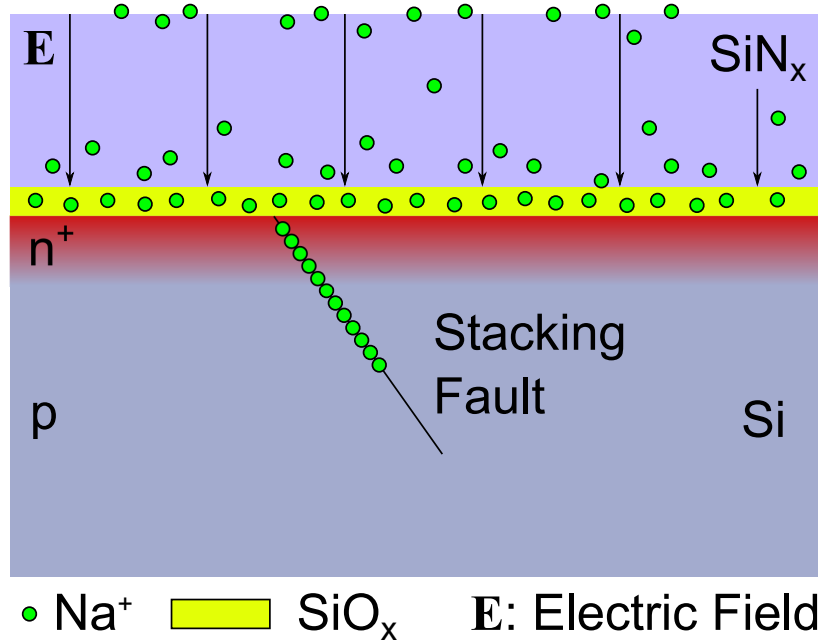


Figure 1.3: Stacking fault decoration schematic illustrating the mechanism behind shunting in p-Si devices. Adapted from Ref. [12]

359 centration of defect states in the band energy diagram, within the p-n junction of a solar cell,  
 360 which induce undesirable recombination. These defects are ultimately responsible for a substan-  
 361 tial decrease in  $R_{sh}$  [9]. The mechanism is illustrated in Fig. 1.3, where it is depicted how ionic  
 362 Na is transported through the  $\text{SiN}_x$  anti-reflective coating of the module, driven by the electric  
 363 field. Then, a large concentration of Na builds up at the interfacial silicon oxide film, followed  
 364 by in-diffusion to the stacking faults in the silicon emitter. Once a critical concentration of Na is  
 365 accumulated in the SF, the defect results in shunting of the junction of the solar cell. Therefore this  
 366 mode of degradation is referred to as PID of the shunting type or PID-s.

### 367 1.3 Origin of Na contamination

368 The front cover glass has been pointed out as the most likely source of Na ions in the mod-  
 369 ule stack [7, 18, 20, 21], mediated by an activation process involving temperature and humidity



370 [21, 22]. Nevertheless, PID-s has been reported to occur in the absence of a front cover glass  
 371 [23] and even in the absence of the polymer encapsulant, when the electric field is supplied by a  
 372 corona discharge [24, 25]. Evidence of Na contamination has been reported within the EVA poly-  
 373 mer encapsulant specially close to the surface of SiN<sub>x</sub> [26]. This suggests that Na contamination  
 374 introduced during the module lamination. Nevertheless, introduction of Na contamination from  
 375 the front cover glass can not be ruled out, especially in combination with high contents of moisture  
 376 and elevated temperatures.

## 377 1.4 Electric Potential in the module

378 The magnitude of the electric field in the module stack and, specifically in the SiN<sub>x</sub> film  
 379 determines to a large extent the ingress of Na in the emitter of the device and hence, susceptibility  
 380 to PID-s. The potential drop across the module stack has been explained by a voltage divider model  
 381 [27]. This model assumes that the main path of current occurs through a circuit of series resistors  
 382 corresponding to the glass, polymer encapsulant, and SiN<sub>x</sub> layers, as shown in Fig. 1.4. Since the  
 383 highly doped emitter (n<sup>+</sup>-Si) has a resistivity in the order of mΩ cm, its contribution can be safely  
 384 neglected. The leakage current  $I_{\text{leak}}$  is given by

$$I_{\text{leak}} = \frac{V_{\text{stress}}}{R_{\text{module}}} = \frac{AV_{\text{stress}}}{\sum_{n=1}^3 \rho_n l_n}, \quad (1.1)$$

385 where  $V_{\text{stress}}$  is the voltage stress applied to the module stack,  $R_{\text{module}} = R_{\text{glass}} + R_{\text{poly}} + R_{\text{SiN}}$ ,  $A$  is  
 386 the area of the device and  $\rho_n, l_n$  are the electrical resistivity and thickness of the  $n$ th-layer (1 →  
 387 glass, 2 → encapsulant, 3 → SiN<sub>x</sub>). Then, the voltage drop  $V_n$  at each material is given by

$$V_n = I_{\text{leak}} R_n = \frac{\rho_n l_n}{\sum_{m=1}^3 \rho_m l_m} V_{\text{stress}} \quad (1.2)$$

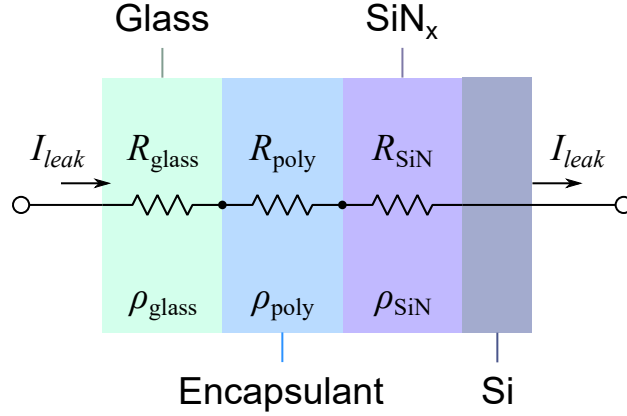


Figure 1.4: Schematic of a voltage divider model corresponding to the stack of materials at the front of Si solar module.

388

## 389 1.5 Sodium Transport

390 While some phenomenological models to describe the kinetics of PID-s have been proposed  
 391 [14, 28, 29], in order to provide adequate solutions to prevent PID-s, a mechanistic understanding  
 392 of the processes behind the shunt formation is required. In that sense, very few attempts have been  
 393 made to propose a physical description of PID-s kinetics [15]. Transport of  $\text{Na}^+$  in the module  
 394 stack is governed by Nernst-Planck equation

$$\frac{\partial C(\mathbf{x}, t)}{\partial t} = -\nabla \cdot \mathbf{J} = -\nabla \cdot [-D\nabla C - \mu C \nabla \phi(\mathbf{x}, t)], \quad (1.3)$$

395 where  $C(\mathbf{x}, t)$  is the concentration of  $\text{Na}^+$  in units of ions per unit volume, as a function of the  
 396 position  $\mathbf{x}$  and time  $t$ ,  $\mathbf{J}$  represents the total flux in units of ions per unit area per time,  $D$  is the  
 397 diffusion coefficient of  $\text{Na}^+$  in the material in units of unit area per time,  $\mu = qD/k_B T$  is the ionic  
 398 mobility in units of  $\text{cm}^2/\text{V}/\text{s}$ ,  $q$  is the elementary charge,  $k_B$  is Boltzmann's constant,  $T$  is the  
 399 absolute temperature and  $\phi(\mathbf{x}, t)$  is the electric potential.

400 We shall see that, due to the discontinuity of the properties (e.g. diffusion coefficient, equi-  
401 librium Na concentration), numerical integration generally is required, except for a very limited  
402 subset of conditions where coarse approximations can simplify the problem [15].

## 403 **1.6 Summary**

404 Engineering solutions to alleviate PID-s requires a mechanistic model to describe the pro-  
405 cesses behind shunt formation. While understanding of the mechanisms leading to loss of per-  
406 formance has remarkably advanced over the past decade, the kinetics of degradation still needs  
407 clarification. Phenomenological models can successfully predict some of the trends and timescales  
408 for PID degradation but fail to correlate material properties that can be used to engineer PID-robust  
409 silicon modules.

410 The motivation of this work is to propose a physical model that mechanistically explains  
411 potential-induced degradation in the solar module. The overarching goal is to parametrize the  
412 kinetics of PID-s and explain the factors that make a module PID-robust.

## 2 | Na transport kinetics in the module stack

The decoration of stacking faults in the n<sup>+</sup>-Si emitter of the PV module follows the transport of Na contamination through the different materials in the module stack. This process is driven by the large voltage differential between the frame and the cell at the extremes of the array. Transport is governed by Nernst-Planck equation (1.3)

$$\frac{\partial C(\mathbf{x}, t)}{\partial t} = \nabla \cdot (D \nabla C) + \mu C \nabla \cdot \nabla \phi(\mathbf{x}, t) + \mu \nabla C \cdot \nabla \phi(\mathbf{x}, t), \quad (2.1)$$

Eqn. (2.1) needs to be solved for each material and coupled with adjacent materials by means of adequate boundary conditions. This is illustrated in Fig. 2.1 where the module stack is schematically drawn alongside with the potential drop due to the applied voltage stress on the layer stack.

The first difficulty in this approach is the fact that the values of the diffusion coefficient of Na in the module stack are either not available for the materials, or are extrapolated from measurements performed at temperatures not relevant to PID [30–32]. It is required that characterization of Na kinetics is performed under conditions as close as possible to operational, or with accelerated testing.

As a first approximation, we assume that Na<sup>+</sup> in the front cover glass is accessible directly at the glass/encapsulant interface in such a way that release is not rate limiting. This is in agreement with the assumption that Na<sup>+</sup> transport is a temperature- and humidity- activated process in soda

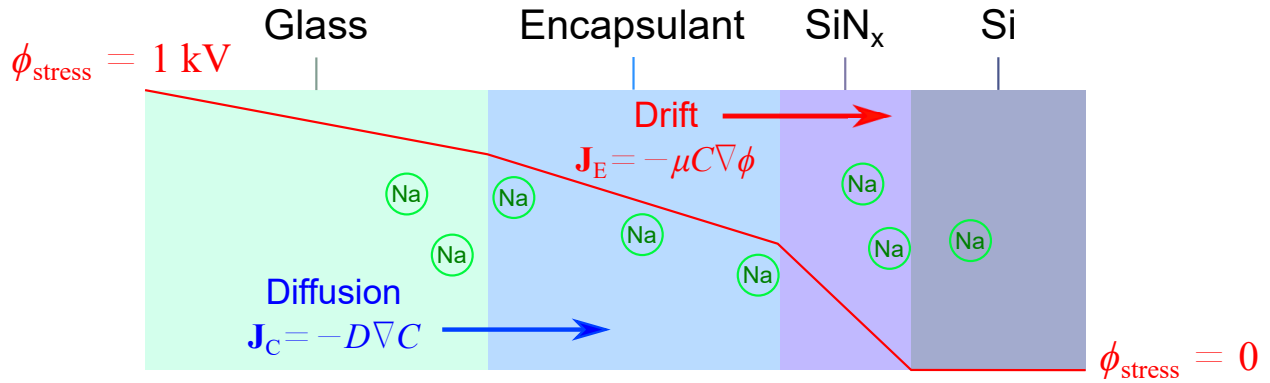


Figure 2.1: Schematic of transport of Na across the different materials in the module stack. According to the voltage divider model, application of a voltage stress over the module leads to a potential drop of different magnitude in each material. The resulting electric field induces a drift term that adds to the otherwise diffusive transport through the materials. The potential in the silicon emitter is neglected due to the large conductivity of this region, hence transport is assumed to occur just by diffusion in Si.

429 lime glass [21, 22]. In other words, we assume that the front cover glass plays the role of a source  
 430 of ions at the glass/encapsulant interface, which reduces the number of materials on which Eq.  
 431 (2.1) needs to be solved.

432 We also assume that the electric field  $\mathbf{E} = -\nabla\phi$  is normal to the glass/encapsulant interface.  
 433 Furthermore, it is be assumed that diffusion can be approximated as a one-dimensional problem  
 434 which simplifies the computation without loss of generality.

## 435 2.1 Approximate voltages in the module stack

436 As discussed in Sec. 1.4, the potential drop in each material depends on the resistivity and  
 437 thickness of all of the other materials in the module stack. Table 2.1 shows typical literature values  
 438 of the resistivities and thicknesses of the different materials in the module stack. The estimated  
 439 electric field is based upon an applied stress voltage of  $V_{\text{stress}} = 1000\text{V}$ . Notice that the real values  
 440 of the resistivities can be subject to orders of magnitude variations depending on the specific pro-

Table 2.1: Typical thicknesses and resistivities of the materials in a c-Si module

Layer	Material	Thickness <sup>e</sup> (cm)	Resistivity ( $\Omega$ cm)	Estimated $E$ (kV/cm) <sup>a</sup>
1	Glass <sup>b</sup>	0.32	$3 \times 10^{12}$	–
2	EVA <sup>c</sup>	$4.5 \times 10^{-2}$	$10^{12} - 10^{15}$	$10^{-1} - 10^4$
3	SiN <sub>x</sub> <sup>d</sup>	$75 \times 10^{-7}$	$10^4 - 10^{19}$	$10^{-2} - 10^5$

<sup>a</sup> Assuming a stress voltage of 1000 V.

<sup>b</sup> Ref. [34]

<sup>c</sup> Ref. [3, 34]

<sup>d</sup> Fit to the ohmic regime in Ref. [27] and Ref. [33].

<sup>e</sup> Ref. [34]

441 cessing conditions of the materials, as well as on the relative humidity and temperature [3, 22, 27].

442 In particular, electronic transport in SiN<sub>x</sub> processed by plasma-enhanced chemical vapor deposi-

443 tion (PECVD) can range between  $10^4$  to  $10^{19}$   $\Omega$ cm depending on the Si/N ratio introduced during

444 deposition [33].

## 445 2.2 Transport of Na in EVA

446 The widespread use of EVA as encapsulant in Si solar modules makes it important to de-

447 termine the impact of Na<sup>+</sup> transport in this polymer. Induced-coupled plasma (ICP) measurements

448 have successfully identified Na contamination in Si modules within the EVA encapsulant, close to

449 the SiN<sub>x</sub> ARC, even prior to PID test [26]. To validate this assumption, Na drift-diffusion exper-

450 iments were performed in metal-EVA-metal structures to determine the penetration depth of Na

451 under conditions as close as possible to the ones referenced in Table 2.1.

### 452 **2.2.1 Experiment**

453 Metal-insulator-metal (MIM) structures were prepared by coating aluminum foil with EVA.  
454 Prior to the coating, aluminum foil is cleaned with acetone, isopropyl alcohol and deionized water.  
455 After curing, the exposed surface of EVA was intentionally contaminated with Na by thermally  
456 evaporating NaCl at  $\sim 800^\circ\text{C}$ . After contamination, silver paste was deposited onto the contami-  
457 nated surface to complete the MIM structure. The thickness of EVA was determined from multiple  
458 point measurement using scanning electron microscopy (SEM). The average value of the thickness  
459 of EVA is  $440\ \mu\text{m}$ .

460 The MIM structures were put on a hot plate at  $80^\circ\text{C}$  and subject to a voltage stress of  $100\ \text{V}$ ,  
461 equivalent to  $2.3\ \text{kV/cm}$  (roughly one third of the value estimated in Table 2.1). Three repetitions  
462 were performed for the same conditions. Relative humidity was not monitored.

463 The samples were sent to the National Renewable Energy Laboratory to perform time-of-  
464 flight secondary-ion mass spectroscopy characterization (ToF-SIMS). After contact removal, the  
465 samples were rinsed in deionized water to remove excess Na. For ToF-SIMS analysis the samples  
466 were sputtered from the side onto which Na contamination was introduced.

### 467 **2.2.2 Results**

468 Fig. 2.2 shows a representative SIMS profile of a sample stressed at  $80^\circ\text{C}$  and  $2.3\ \text{kV/cm}$   
469 for 12 h. Due to the large concentration of ions at the interface, a large memory effect is observed  
470 in the concentration profile, due to re-deposition of Na during sputtering. During this timescale, Na  
471 reached a depth of  $\sim 15\ \mu\text{m}$ , beyond which, the concentration drops below detectable limits. By fit-  
472 ting the concentration profile to Eq. (3.7), we estimated a diffusion coefficient of  $4 \times 10^{-14}\ \text{cm}^2/\text{s}$ .

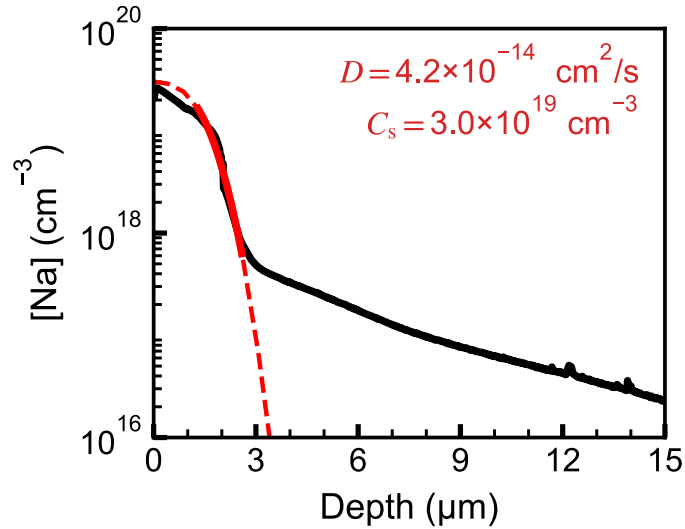


Figure 2.2: Representative Na drift-diffusion profile obtained by SIMS of a metal-EVA-metal structure intentionally contaminated with Na on one of the ends stressed for 12 h under an electric field of 2.3 kV/cm at a temperature of 80 °C.

473 Using Eq. (3.9), we determined that the characteristic Na transit time, for a 300 μm thick EVA  
 474 encapsulant stressed at 7 kV cm<sup>-1</sup> is ≈ 36 h. This finding contrasts with the trends on the nor-  
 475 malized power output observed on PID-susceptible devices from reports of p-Si devices stressed  
 476 under similar conditions [5, 35–38], and those from normalized  $R_{sh}$  [24, 39–41] for which a 5 %  
 477 degradation of the initial condition occurs within an average time of ≈ 13 h. While Eq. (3.7) does  
 478 not consider the contribution of the ionic concentration to the electric potential, the net effect of  
 479 this is to screen the applied electric stress. This means that, when considering the image charge  
 480 effect of the ion distribution, drift decreases considerably close to the source. This result indicates  
 481 that transport in EVA has a limited impact in the kinetics of PID. Given this result, we think that  
 482 transport of Na to the cell interface occurs during lamination, or that Na is already present on cell



## 483 **2.3 Summary**

484 The typical electrical stress in each material of the PV module was estimated using a volt-  
485 age divider model, for which a potential drop ranging from  $10^{-2}$  V cm<sup>-1</sup> to  $10^4$  V cm in the SiN<sub>x</sub>  
486 film was estimated. Drift-diffusion experiments in EVA showed that Na<sup>+</sup> is not expected to tra-  
487 verse the full depth of the encapsulant at significant concentrations to cause PID-s in the timescales  
488 within which degradation is typically reported on PID-prone devices. This finding is in agreement  
489 with reports suggesting the presence of Na<sup>+</sup> on the surface of SiN<sub>x</sub> prior to PID tests. It is pos-  
490 sible that damp-heat tests at a high relative humidity (85 % RH) lead to electrochemical reactions  
491 that are not accounted for in the scope of this work. Nevertheless, changes in the conductivity of  
492 EVA associated to different levels of water content and temperature might have an impact on the  
493 magnitude of the electric field in SiN<sub>x</sub>.

### 3 | Transport of Na in SiN<sub>x</sub>

Sodium moves through SiN<sub>x</sub> as an ionic species subject to the kinetics described by Eq. (3.7). In this chapter we show that transport in SiN<sub>x</sub> is very sensitive to the magnitude of the applied electric potential, as well as the average concentration of Na<sup>+</sup>. We also establish a number of conditions within which, we can simplify Eq. (3.7), in order to derive analytical approximations that allow us to characterize the drift-diffusion kinetics of Na contamination in dielectrics. While some reports suggest that SiN<sub>x</sub> films of the sort used in solar modules can act as barriers against mobile charge contamination [42–44], there is also evidence that, under an electric field of 0.5 MV/cm, sodium can drift across an 80 nm thick layer of SiN<sub>x</sub> in just 25 min at 80 °C [32]. Given the large variations in the resistivities of the different materials in p-Si modules, electric fields of the order of 0.5 MV/cm can't be discarded. Therefore, accurate knowledge of Na kinetics in silicon nitride under typical module operating conditions is necessary.

## 3.1 Characterization of drift-diffusion kinetics using C-V measurements

Mobility of ionic charge in dielectrics has been addressed in the past by means of transient current measurements [45–48]. This method assumes that mobile charges drift over the entire length of the dielectric during the measurement. This assumption is adequate for ions with mobilities fast enough to enable drift across the dielectric within the time scale of the experiment or less [48]. Nevertheless, ions with lower diffusivities might not satisfy this criterion, especially at low temperatures. In contrast, methods relying on the determination of the flatband voltage ( $V_{FB}$ ) from capacitance-voltage (C-V) methods, enable the detection of ion migration in dielectrics on a broader temperature range. Among these methods, the bias-temperature-stress (BTS) method has been widely used to determine ionic migration [43–45, 49–52].

## 3.2 Trap-corrected bias-temperature-stress method

In the standard BTS method, ionic contribution to flatband voltage measurements can be obfuscated by carrier trapping effects and hence, modifications are required to adequately quantify ion kinetics. A signature due to trapping effects has in fact been identified in dielectrics that are known to have a large density of trap centers, including  $\text{SiN}_x$  [49, 53, 54] and high- $k$  dielectrics such as hafnium oxide [55]. Conflation of flatband shift due to carrier trapping in ionic mobility measurements results in incorrect attribution of drift characteristics as seen in porous  $\text{SiO}_x$  [51], which precludes reliable analysis of migration in the dielectric. We developed a time-resolved

525 capacitance-voltage method which incorporates a kinetics model to estimate the ionic diffusivity  
 526 in dielectrics. We used this method to quantify the diffusion coefficient of Na<sup>+</sup> in SiN<sub>x</sub> which is  
 527 necessary to study the ingress of Na in the emitter of p-Si modules leading to PID.

### 528 **3.2.1 Flatband fitting**

529 In the BTS method, the barrier for ionic diffusion is lowered by providing a thermal stress  
 530 (i.e. heating the device above room temperature) and an electrical stress, by applying an electro-  
 531 static potential  $V_{\text{stress}}$  to the dielectric. The flatband voltage at time  $t$  with respect to its value at  $t = 0$   
 532 is determined by integration of Poisson's equation in the dielectric

$$\frac{d^2\phi}{dx^2} = -q \frac{n(x,t)}{\epsilon_{\text{ins}}}, \quad n(x,t) \equiv z_{\text{ion}}C(x,t) + z_{\text{traps}}n_{\text{traps}}(x,t), \quad (3.1)$$

where  $z_{\text{ion}}$ ,  $z_{\text{traps}}$  are the valency of ionic species and traps respectively ( $z_{\text{ion}} = 1$ , for Na<sup>+</sup> and  
 $z_{\text{traps}} = \pm 1$ , depending on the type of carrier captured),  $C(x,t)$ ,  $n_{\text{traps}}(x,t)$  are the concentration  
 of mobile ions and occupied traps respectively, and  $\epsilon_{\text{ins}}$  is the permittivity of the dielectric. Eq.  
 (3.1) is integrated from a position in the insulator to the position  $x = L$ , located at the surface of  
 the semiconductor. The boundary conditions assumed are that the band bending and the potential  
 vanish at the silicon surface in the flatband condition ( $d\phi/dx \rightarrow 0$  and  $\phi \rightarrow 0$  as  $x \rightarrow L$ ) and, that  
 $\phi(x=0) = V_{\text{FB}} - W_{\text{ms}}$ ; where  $W_{\text{ms}}$  represents the difference between the work function of the metal  
 and that of the semiconductor. This leads to the general expression for  $V_{\text{FB}}$  [56]:

$$V_{\text{FB}}(t) = W_{\text{ms}} - \frac{q}{\epsilon_{\text{ins}}} \int_0^L x' n(x',t) dx'. \quad (3.2)$$

533 Assuming that the concentration of  $\text{Na}^+$  and occupied traps is negligible at  $t = 0$ , we can define the  
 534 shift in flatband voltage as

$$\Delta V_{\text{FB}}(t) \equiv V_{\text{FB}}(t) - V_{\text{FB}}(t = 0) = - \underbrace{\frac{q}{\epsilon_{\text{ins}}} \int_0^L x' C(x', t) dx'}_{\Delta V_{\text{FB}}^{\text{ion}}(t)} - \overbrace{\frac{q z_{\text{traps}}}{\epsilon_{\text{ins}}} \int_0^L x' n_{\text{traps}}(x', t) dx'}^{\Delta V_{\text{FB}}^{\text{traps}}(t)}, \quad (3.3)$$

535 which is independent of  $W_{\text{ms}}$  and captures the effect of occupied traps in the dielectric.

536 Eq. (3.3) also implies that, if ions and traps do not interact, their contribution can be  
 537 decoupled which is the basis of the trapping correction.

538 The ionic contribution to the flatband voltage shift  $\Delta V_{\text{FB}}^{\text{ion}}(t)$  can be now estimated by sub-  
 539 tracting the component due to electronic trapping  $\Delta V_{\text{FB}}^{\text{traps}}(t)$ , measured on devices without inten-  
 540 tional ion contamination  $C(x, t) = 0$ , from the flatband voltage shift measured on ion-contaminated  
 541 devices:

$$\Delta V_{\text{FB}}^{\text{ion}}(t) = \Delta V_{\text{FB}}(t) - \Delta V_{\text{FB}}^{\text{traps}}(t). \quad (3.4)$$

### 542 **3.3 Kinetics of ion diffusion and drift in $\text{SiN}_x$**

543 The kinetics of ion transport is then obtained by relating  $\Delta V_{\text{FB}}^{\text{ion}}(t)$  to the ionic concentration  
 544 as prescribed by Eq. (2.1). From Eq. (3.1), we see that the electric potential is a function of the  
 545 ionic and trap concentrations in the dielectric itself, which implies that Nernst-Planck and Poisson's  
 546 equations are coupled. Nevertheless, at times less than a critical value  $\tau_c$ , beyond which accumu-  
 547 lation of ions at the interface leads to surface charges induce  $E = Q/\epsilon$  comparable with  $V_{\text{stress}}/L$ ,  
 548 the concentration can assume a limiting constant value  $C(x) = C$ . In such a case, the solution to

549 the ionic component of (3.1) is simply  $\phi = -(q/\epsilon_{\text{ins}})L^2C$ . This means that for concentrations

$$C \ll C_{\text{th}} \equiv \frac{2\epsilon_{\text{ins}}}{qL^2}V_{\text{stress}} \quad (3.5)$$

550 the magnitude of the ionic charge distribution is much less than the applied electric potential, and

551 the electric field can be assumed to be  $V_{\text{stress}}/L$ . The Na-contaminated  $\text{SiN}_x$  films prepared for this

552 work were analyzed using X-ray photoelectron spectroscopy (XPS) which has a detection limit of

553  $5 \times 10^{18} \text{ cm}^{-3}$  and no signal corresponding to Na was detected. In comparison, the magnitude of

554 the applied voltage is 8.6 V over a 86 nm thick layer of  $\text{SiN}_x$  ( $\epsilon_r \approx 7$ ), which sets the right hand side

555 of (3.5) on the order of  $10^{18} \text{ cm}^{-3}$ , indicating that, indeed  $C$  is low compared to  $V_{\text{stress}}$  and hence,

556  $E = -\nabla\phi \rightarrow V_{\text{stress}}/L$ . This allows us to assume a constant drift velocity  $v_d = -\mu\partial_x\phi = zqDE/k_B T$

557 and remove Laplacian term in (2.1):

$$\frac{\partial C(x,t)}{\partial t} = \frac{\partial}{\partial x} \left( D \frac{\partial C}{\partial x} \right) - v_d \frac{\partial C}{\partial x}, \quad (3.6)$$

558 where we have reduced the problem to one-dimension. The advection term in (3.6) can be elimi-

559 nated by introducing the moving reference frame [57, 58]

$$x' = x - v_d t, \quad t' = t,$$

560 which transforms Eq. (3.6) to a pure diffusion equation in the primed reference frame. We assume

561 a constant source of ions at the surface of  $\text{SiN}_x$ :  $C(x=0,t) = C_s$  and a closed boundary at the

562 Si interface:  $[-D\partial_x C + v_d C]_{x=L} = 0$ . Given these assumptions, we can now define the analytical

563 solution of Eq. (3.6) for this set of boundary conditions, before the onset of accumulation ( $t < \tau_c$ ,

564 discussed in Sec. 3.6) as follows

$$C(x,t) = \frac{C_s}{2B} \left[ \text{erfc} \left( \frac{x - \mu E t}{2\sqrt{Dt}} \right) + \text{erfc} \left( -\frac{x - 2L + \mu E t}{2\sqrt{Dt}} \right) \right], \quad (3.7)$$

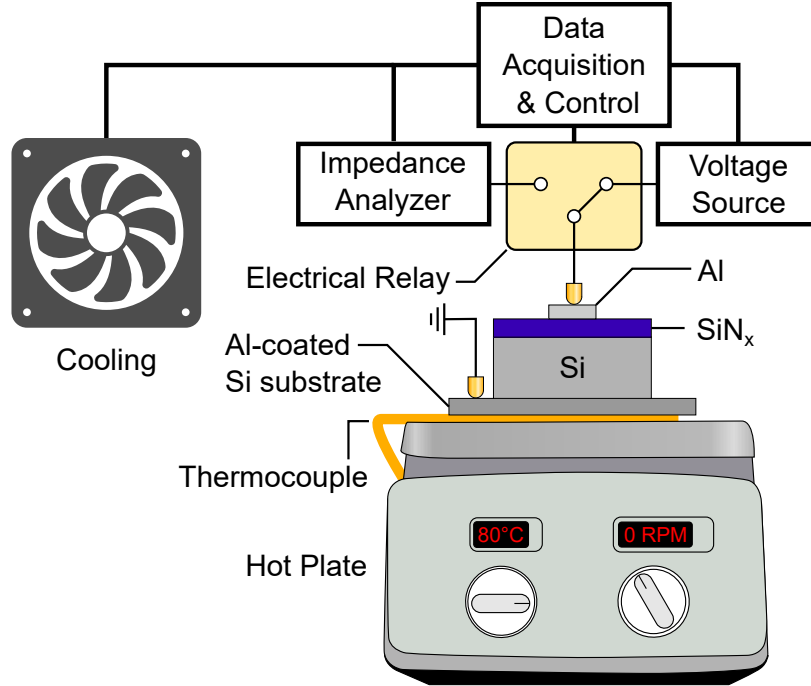


Figure 3.1: Schematic of the BTS setup used to quantify Na<sup>+</sup> migration in SiN<sub>x</sub>.

565 with a normalization constant  $B \equiv \operatorname{erfc}(-\mu Et/2\sqrt{Dt})$ .

566 The diffusivity of ionic species  $D$  and concentration of ions at the source  $C_s$  can be deter-  
 567 mined from the experimental  $\Delta V_{\text{FB}}$  by fitting the theoretical flatband shift obtained from (3.3) and  
 568 (3.7) to the experimental flatband shift obtained through (3.4).

569 The fitting is performed using trust-region reflective nonlinear least squares. The uncer-  
 570 tainty of the fitted parameters is taken from the 95 % confidence intervals estimated from the co-  
 571 variance matrix of the fit and using the  $t$ -student distribution.

## 572 3.4 Experiment

573 Metal-insulator-semiconductor (MIS) structures were fabricated on  $n$ -type silicon (100)  
 574 wafers with a resistivity of  $2.7 \Omega \text{ cm}$ . The wafers were cleaned with acetone, ethanol and isopropyl

575 alcohol and then cleaned with RCAI and RCAII procedures, followed by a 1 % hydrofluoric acid  
576 dip.  $\text{SiN}_x$  films were grown on the polished side of the wafer using PECVD with a  $\text{NH}_3/\text{SiH}_4$  ratio  
577 of 7. This ratio was chosen to obtain high  $\text{SiN}_x$  resistivity [59] to ensure a significant potential drop  
578 across the dielectric, which leads to drift times within reasonable timescales in the experiments.  
579 Additionally,  $\text{SiO}_2$  films were grown by dry oxidation at a temperature of 1100 °C. The thickness  
580 and refractive index of the as-grown films were mapped using spectroscopic ellipsometry. An  
581 average thickness of 86 nm and an average index of refraction of 1.89 at 600 nm were estimated  
582 for the  $\text{SiN}_x$ . The dielectric layers were subsequently cleaned with isopropyl alcohol and ethanol to  
583 remove surface contaminants and the back surface was swabbed with hydrofluoric acid. Aluminum  
584 gate electrodes of approximately 200 nm were deposited by thermal evaporation onto  $\text{SiN}_x$ . The  
585 electrodes were defined by a shadow mask with 1 mm diameter circles and a pitch of 2.54 mm.  
586 A layer of aluminum was deposited on the bottom side of the wafer to produce the back contact.  
587 In devices with intentional contamination, 25 nm of NaCl were thermally evaporated at 800 °C,  
588 followed by the deposition of the gate without breaking vacuum.

589 An in-situ C-V setup with automated BTS capabilities was developed to perform migration  
590 studies. As illustrated in Fig. 3.1, the temperature stress is provided by a temperature-controlled  
591 hotplate with feedback from a thermocouple directly placed underneath the sample. A software-  
592 controlled relay was used to switch between stressing and measuring modes in order to periodically  
593 monitor  $\Delta V_{\text{FB}}$  as a function of stress BTS time. Contact to the back of the sample was achieved by  
594 placing the sample on an Al-coated silicon wafer, which provides adequate thermal distribution.  
595 Contact to the Al-coated wafer and the gate was achieved by means of Au-plated spring-loaded  
596 contacts. The system applied BTS stress to the MIS capacitors in 1 or 4 h intervals using a pro-  
597 grammable hotplate and a Keithley 2401 source meter to apply a direct current (DC) bias. It has



598 been reported that an electric field on the order of  $1 \text{ MV cm}^{-1}$  is required to observed PID in sil-  
599 icon PV modules, which contrasts with the voltage-divider estimation in Table 2.1. The reason  
600 for this discrepancy is unknown [60], but it could be attributed to large variations in the electrical  
601 properties introduced during the PECVD deposition of  $\text{SiN}_x$ . An electric field of  $1 \text{ MV cm}^{-1}$ , is  
602 adequate for our experiments since it satisfies Eq. (3.5) and was thus applied to the MIS structures  
603 during migration experiments, in agreement with other studies [32]. After each BTS interval, a fan  
604 (Fig. 3.1) was used to cool down the devices to room temperature within 15 min. Once cooled,  
605 C-V curves were acquired with a HP4194A impedance analyzer at 1 MHz, to ensure that ions  
606 were unaffected by the 100 mV (peak to peak) alternating current (AC) signal from the impedance  
607 analyzer. During the C-V acquisition, the DC voltage was swept from  $-8.6$  to  $8.6 \text{ V}$  and each  
608 measurement was averaged 32 times with a dwell time of 5 ms per point. The flatband voltage  
609 as a function of BTS time was estimated from the C-V curves using a second-derivative method  
610 described in Ref. [61].

## 611 **3.5 Results**

612 The need to correct for traps in the dielectric to obtain the ionic component to  $\Delta V_{\text{FB}}$  is  
613 shown in Fig. 3.2. In devices without intentional Na contamination,  $\Delta V_{\text{FB}}$  increases as a function  
614 of time, which is explained by an increase in negative charge in the dielectric according to (3.3).  
615 This effect originates from electron trapping as demonstrated in a previous study [54], and can  
616 be further quantified by integrating Frenkel-Poole emission current [63]. A first-order kinetics  
617 model [64] was used to determine a trapping time constant of 2.5 h. Because each C-V sweep  
618 is completed in 2 min, it is not expected that the DC bias from the C-V measurement influences

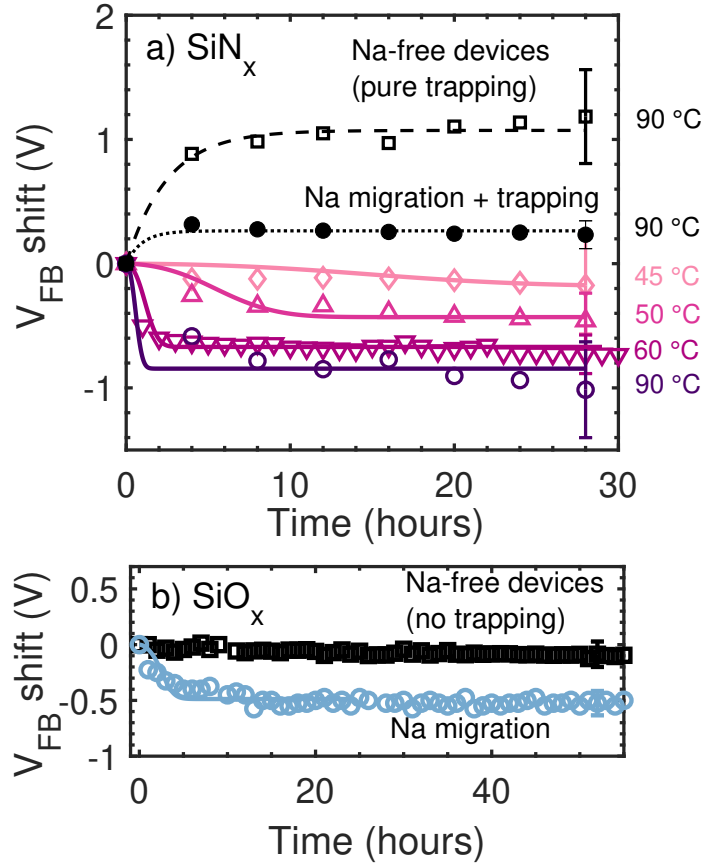


Figure 3.2: Flatband shift as a function of time and temperature: (a) in  $\text{SiN}_x$  capacitors stressed at  $1 \text{ MV cm}^{-1}$ . An example of raw flatband shift including both  $\text{Na}^+$  and trapping is shown in black empty squares with a dotted line as a guide to the eye. Trapping-corrected flatband voltage shifts are shown at different temperatures. The fits obtained from the Nernst-Planck migration model are depicted by solid lines. (b)  $\text{SiO}_2$  capacitors without trapping correction (pure ionic shift) in both Na-free and Na-contaminated devices at  $0.5 \text{ MV cm}^{-1}$  and  $50 \text{ }^\circ\text{C}$ . Reproduced from Ref. [62]

619 the flatband shift. The raw  $\Delta V_{\text{FB}}$  signal (before trapping correction) in a Na-contaminated sample  
 620 is shown in solid black circles in Fig. 3.2, which includes the contributions from both, electron  
 621 trapping (negative charging) and Na migration (positive charging). To account for charge due  
 622 electron trapping in  $\text{SiN}_x$  [49, 53, 54], we estimated  $\Delta V_{\text{FB}}^{\text{traps}}$  by averaging the flatband shift from  
 623 four Na-free devices and subtracted its contribution from  $\Delta V_{\text{FB}}$ , as prescribed by Eq. (3.4). After  
 624 applying this correction, a negative flatband shift is observed as a function of time, as expected from  
 625 the ingress of positive ions into the dielectric. Note that the error bars originate from variations in

626 the concentration of trapped charge in Na-free devices used in the estimation of  $\Delta V_{\text{FB}}^{\text{traps}}$ . Overall,  
627 our trapping-corrected data captures the resulting  $\Delta V_{\text{FB}}$  in Na-contaminated devices across device-  
628 relevant temperatures.

629 To confirm the applicability of the trapping-corrected BTS method to isolate ion migration,  
630 experiments were conducted on thermal  $\text{SiO}_2$  capacitors, which are known to exhibit very little  
631 to no carrier trapping in the bulk of the dielectric compared to  $\text{SiN}_x$  [49, 63, 65]. Fig. 3.2 shows  
632 the measured  $\Delta V_{\text{FB}}$  without trapping correction in Na-free and Na-contaminated  $\text{SiO}_2$  capacitors.  
633 As expected, we observed little-to-no variation in the flatband voltage in control samples without  
634 intentional Na contamination. In Na-contaminated  $\text{SiO}_2$  capacitors, a flatband shift of  $\approx 0.5$  V was  
635 detected without the need for trapping correction. This trend is well fit by our ion migration model.  
636 This result confirms that the experimental conditions and detection method are suitable to detect  
637 ion migration within the timescales of the experiment.

638 Using the described trapping-corrected BTS method, we determined the Arrhenius relation-  
639 ship for the migration of  $\text{Na}^+$  in  $\text{SiN}_x$  by averaging the diffusivities of measurements collected over  
640 4–8 devices at each temperature. Fig. 3.3 shows the extracted diffusivities of  $\text{Na}^+$  in  $\text{SiN}_x$  at differ-  
641 ent temperatures together with the 95 % confidence interval. The results of the fit to the Arrhenius  
642 relationship give a diffusion coefficient prefactor of  $D_0 = 1.4 \times 10^{-14} \text{ cm}^2/\text{s}$  with a 95 % confi-  
643 dence interval of  $[1.2 \times 10^{-15}, 1.6 \times 10^{-13}] \text{ cm}^2/\text{s}$  and an activation energy of  $E_a = 0.14 \text{ eV}$ , with  
644 a 95 % confidence interval of  $[0.07, 0.21] \text{ eV}$ . Note that the deviation from the higher-temperature  
645 results of Wilson et al. [32] indicates a large difference in the activation energy compared to the  
646 range of temperatures in this work.

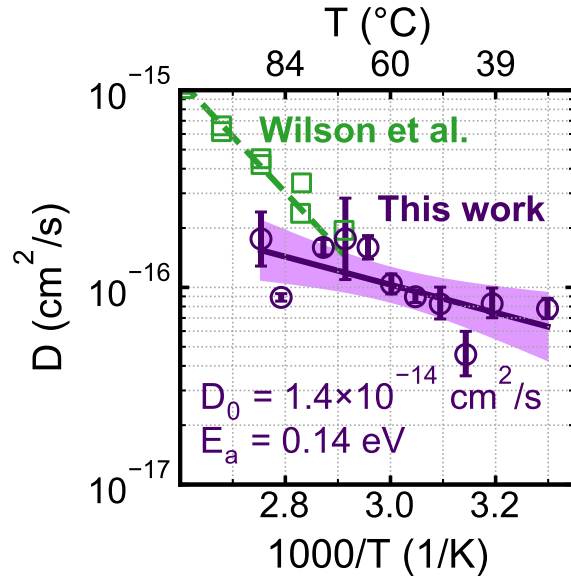


Figure 3.3: Arrhenius relationship for the diffusivity of  $\text{Na}^+$  in  $\text{SiN}_x$  evaluated by the trapping-corrected BTS method and by Wilson et al. [32], using a Kelvin-probe-based method. The estimated prefactor for the diffusion coefficient is  $D_0 = 1.4 \times 10^{-14} \text{ cm}^2/\text{s}$  with a 95 % confidence interval of  $[1.2 \times 10^{-15}, 1.6 \times 10^{-13}] \text{ cm}^2/\text{s}$ . The estimated activation energy is  $E_a = 0.14 \text{ eV}$ , with a 95 % confidence interval of  $[0.07, 0.21] \text{ eV}$ .

### 647 3.6 Characteristic $\text{Na}^+$ transit time

648 To provide context to the results of the ion-migration experiments with respect to PID-s, we  
 649 estimate how the  $\text{Na}^+$  concentration at the  $\text{SiN}_x/\text{Si}$  interface changes over time. Since Na transport  
 650 can eventually lead to PID-s, we consider the characteristic time  $\tau_c$  at which  $\text{Na}^+$  drifts across the  
 651 dielectric, as the solution to the equation

$$L = 2\sqrt{D\tau_c} + \mu E \tau_c \quad (3.8)$$

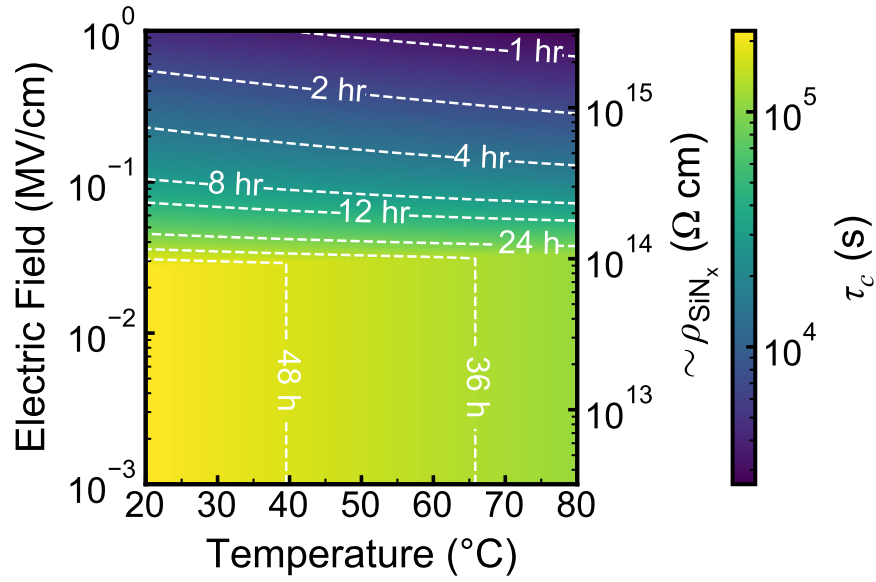


Figure 3.4: Characteristic Na transit time in a 65 nm-thick  $\text{SiN}_x$  film as a function of the module temperature and electric field across the dielectric.

If the magnitude of the electric field in  $\text{SiN}_x$  is low with respect to the diffusive term,  $-D\nabla^2 C \ll \mu EC$ , then  $\tau_c \rightarrow L^2/D$ . If the drift term dominates  $\tau_c \rightarrow L/\mu E$ . This can be expressed as

$$\tau_c = \inf \left\{ t = \frac{L^2}{D}, \quad t = \frac{L}{D} \left( \frac{v_{\text{th}}}{E} \right) + \frac{2}{D} \left( \frac{v_{\text{th}}}{E} \right)^2 \left\{ 1 \pm \left[ 1 + \left( \frac{E}{v_{\text{th}}} \right) L \right]^{1/2} \right\} \middle| t \in \mathbb{R}^+ \right\}, \quad (3.9)$$

652 where  $v_{\text{th}} = k_{\text{B}}T/q$  is the thermal voltage. The complete derivation of Eq. (3.9) is shown in  
 653 Appendix A.

654 Fig. 3.4 shows the characteristic transit time across a typical nitride thickness of 65 nm as  
 655 a function of the module temperature and electric field across the  $\text{SiN}_x$  in ranges relevant to solar  
 656 module operation. The range of values selected for the electric field in  $\text{SiN}_x$  account for a large  
 657 variability on the resistivities of the materials in the module [3, 12, 22, 33, 34, 66]. To estimate  
 658 the values on the resistivity axis in Fig. 3.4 we consider a borosilicate glass with a resistivity  
 659  $\rho_{\text{glass}} \approx 3 \times 10^{12} \Omega \text{ cm}$  [67], ethylene-vinyl acetate EVA with a resistivity  $\rho_{\text{EVA}} \approx 5 \times 10^{13} \Omega \text{ cm}$   
 660 and a voltage differential of 1000 V, to estimate the silicon nitride resistivity corresponding to

661 the electric field in the left axis. Under such conditions, Na migration through the dielectric can  
662 occur between a few hours to a few days depending primarily on the potential drop across SiN<sub>x</sub>.  
663 Such times are lower than the observed degradation times after accelerated experiments at high  
664 temperature and in the field [38, 68]. However,  $\tau_c$  only reflects the transport across SiN<sub>x</sub> and does  
665 not represent the actual time to PID in the full module stack.

666 It is worth considering scenarios in which the assumptions made to derive Eq. (3.7) do  
667 not longer hold, namely (1) large ion concentrations  $C_s \approx 2\epsilon_{\text{ins}}V_{\text{stress}}/qL^2$  where the distribution of  
668 mobile ions does affect the local potential within the dielectric, (2) long stress times:  $t > \tau_c$  where  
669 the flatband potential becomes sensitive to the accumulation of ions at the dielectric/semiconductor  
670 interface, and (3) finite sources of ions.

### 671 **3.7 Conclusions**

672 We derived a method to quantify Na<sup>+</sup> diffusion kinetics in SiN<sub>x</sub> using a drift-diffusion  
673 model to analyze the corresponding flatband voltage shift in MIS structures. We developed a C-V  
674 setup for measuring MIS capacitors under BTS. We developed a model to correct nonidealities  
675 from charge trapping in the dielectric. This description of a C-V based approach to parametrize  
676 ion migration should allow reliable quantification of ion diffusion kinetics broadly in dielectrics.  
677 We presented an analytical approximation to bound the transport times of Na<sup>+</sup> in SiN<sub>x</sub> within the  
678 context of PID. We showed that migration through PV-relevant, 65 nm-thick silicon nitride films  
679 can occur within a few days in field-operating conditions. We can easily extend this study to a  
680 variety of SiN<sub>x</sub> of different compositions and electrical properties. Furthermore, quantification of  
681 the transport kinetics from this study is especially significant to determine the overall kinetics of

682 PID-s in the full module stack.

683 Part of Chapter 3 has been published in *Phys. Status Solidi A*. under the title “Quantifi-  
684 cation of sodium ion migration in SiN<sub>x</sub> by flatband-potential monitoring at device operating tem-  
685 peratures”, by G. von Gastrow, **E. Martínez-Loran**, J. Scharf, J. Clenney, R. Meier, P. Bandaru,  
686 M. Bertoni, and D. Fenning. The dissertation/thesis author was an author of equal contribution,  
687 together with G. von Gastrow and J. Scharf.

## 4 | Finite Element Model of Mobile-Ion Kinetics

Sodium contamination is known to be deleterious to semiconductor devices since the early days of the semiconductor industry. The need to mitigate this problem opened the way for studying mobile charge transport in dielectrics commonly used in the industry [42–45, 69–73]. In silicon photovoltaic (PV) modules, Na contamination has been determined as the root cause for PID-s, and linked to failure within a timescale of a few hours to a few days under accelerated testing [5, 24, 35, 36, 38, 39, 41, 74]. In section 3.3 we described a simplified model to estimate the kinetics of  $\text{Na}^+$  migration under the limiting case of short migration times ( $t < \tau_c$ ) and low ionic concentrations with respect to the applied electric field ( $C \ll 2\epsilon_{\text{ins}}/qL^2$ ). In this chapter, as general solution to Eq. (2.1) is presented, using only a limited set of assumptions. We then expand the model to accommodate for a flux of ions out of the dielectric and into the semiconductor. This is useful in modeling PID-s kinetics on p-Si modules.



## 701 **4.1 Modeling mobile charge kinetics in MIS structures**

702 The conditions assumed to derive an analytical approximation to Eq. (3.6) are generally not  
703 held for long-time experiments nor for a wider range of mobile charge concentrations, as would  
704 be expected for PID-s in the case of Si PV modules [23, 75, 76]. In metal-insulator-semiconductor  
705 (MIS) structures, the concentration of ions near the metal gate induces an image charge in the  
706 metal, with the opposite sign of the ion. The resulting electric field points in the negative direction  
707 of  $x$ , leading to a diffusion dominated region in the vicinity of the metal gate [69]. It is also typi-  
708 cal to assume a closed boundary at the dielectric/semiconductor interface [56, 69], which leads to  
709 ion accumulation at that interface due to ion drift towards the semiconductor. Just as in the metal  
710 gate, accumulation of ions at the dielectric/semiconductor interface induces an image charge of  
711 the opposite sign in the semiconductor –which we measure as  $\Delta V_{\text{FB}}^{\text{ion}}$  in Eq. (3.4), but in this case  
712 the resulting image charge acts on the same direction of the applied electric field. This effect was  
713 first reported by Snow et al. [69], where the authors introduced a boundary layer approximation to  
714 estimate the image charge in the semiconductor. This approach has the advantage of avoiding the  
715 need to solve the coupled Poisson-Nernst-Planck (PNP) system of equations. Nevertheless, this  
716 model does not mechanistically account for the image charge effect at the source and semiconduc-  
717 tor interfaces or the depletion of ions near the gate. In such cases the Poisson’s equation for the ion  
718 distribution:

$$\nabla \cdot \nabla \phi_{\text{ion}} = -\frac{zqC(x,t)}{\epsilon_{\text{ins}}} \quad (4.1)$$

719 needs to be solved self-consistently with (2.1). Analytical solutions to Poisson-Nernst-Planck  
720 equations are available just for a few limiting cases, namely, low applied voltages and binary

721 electrolytes close to charge neutrality [77–80]. In this chapter, we present a finite-element-method  
 722 (FEM) solution to the Poisson-Nernst-Planck coupled system of equations. This methodology has  
 723 the advantage of fully describing the kinetics of ion migration in MIS structures and providing a  
 724 framework to model transport across stacks of materials. This is specially useful, for instance to  
 725 describe ion ingress leading to shunt formation in PID-s.

726 The boundary conditions for Eq. (4.1) can be set by estimating the image charges induced  
 727 by the ion distribution. The contribution of image charge of ionic origin  $Q_{\text{ion}} = q \int_0^L C(x', t) dx'$  to  
 728 the drift term in Eq. (2.1) arises from  $\text{Na}^+$  charge being imaged at both, the metal gate  $Q_G$  and  
 729 the semiconductor  $Q_S$  interfaces of the MIS structure. The surface charge distribution of image  
 730 charges due to ionic migration are determined by [56, 69]:

$$Q_G = q \int_0^L \left( \frac{x' - L}{L} \right) C(x', t) dx' \leq 0 \quad (4.2a)$$

$$Q_S = -q \int_0^L \left( \frac{x'}{L} \right) C(x', t) dx' \leq 0, \quad (4.2b)$$

731 with  $L$  the thickness of the dielectric. It follows that  $-Q_{\text{ion}} = Q_G + Q_S$ . The electric fields  
 732 associated with these charge distributions have opposite directions inside the dielectric:

$$-\nabla \phi_{\text{ion}}|_{x=0} = \mathbf{E}_{\text{ion,G}} = \hat{\mathbf{x}} \frac{q}{\epsilon_{\text{ins}}} \int_0^L \left( \frac{x' - L}{L} \right) C(x', t) dx' \quad (4.3a)$$

$$-\nabla \phi_{\text{ion}}|_{x=L} = \mathbf{E}_{\text{ion,S}} = -\hat{\mathbf{x}} \frac{q}{\epsilon_{\text{ins}}} \int_0^L \left( \frac{x'}{L} \right) C(x', t) dx' \quad (4.3b)$$

733

734 Thus, for  $\mathbf{E} = E\hat{\mathbf{x}}$ , the image charge field at the gate opposes the external electric field,  
 735 while the image charge field at the semiconductor interface adds to it. Eq. (4.3) can be introduced

736 as the boundary conditions to Poisson's equation

$$\nabla\phi(x=0) \cdot \hat{\mathbf{n}}_G = E_{\text{stress}} + \frac{q}{\epsilon_{\text{ins}}} \int_0^L \left( \frac{x' - L}{L} \right) C(x', t) dx' \quad (4.4a)$$

$$\nabla\phi(x=L) \cdot \hat{\mathbf{n}}_S = -E_{\text{stress}} - \frac{q}{\epsilon_{\text{ins}}} \int_0^L \left( \frac{x'}{L} \right) C(x', t) dx', \quad (4.4b)$$

737 where  $\hat{\mathbf{n}}_G = -\hat{\mathbf{x}}$  and  $\hat{\mathbf{n}}_S = \hat{\mathbf{x}}$  are the normal vectors to the gate and semiconductor surfaces enclosing  
738 the insulator volume in the MIS structure.

## 739 **4.2 Finite Element Method Implementation of the PNP cou-** 740 **pled system**

741 The general solution to the Poisson-Nernst-Planck system requires integration over the  
742 space and time. Spatial integration is relatively straightforward using the finite element method.  
743 The general approach to impose the boundary conditions require writing the spatial part of the  
744 partial differential equation (PDE) as an integral equation (i.e. the “weak form”), which has the  
745 advantage of reducing the computational load of evaluating second order spatial derivatives [81,  
746 82]. This formulation also allows us to easily introduce both, the Dirichlet and Neumann bound-  
747 ary conditions [81]. Time integration must be addressed by a finite difference scheme using the  
748 solutions computed by the FEM solver.

749 Among the different time integration schemes, the two-step trapezoidal-second order back-  
750 ward difference (TR-BDF2) method has proven to be very reliable for semiconductor modeling  
751 [83, 84] because of its strong stability [83–86] compared to other time stepping approaches like  
752 the Crank-Nicholson method, which is known to introduce spurious oscillations if un-damped [87].

753 The TR-BDF2 advances the system

$$\frac{\partial C}{\partial t} = F[C(t), \phi(t), t] \quad (4.5)$$

754 from  $t_n$  to  $t_{n+1} = t_n + \Delta t_n$  in two steps [83, 84]. In Eq. (4.5),  $F$  corresponds in this case to the right  
 755 hand side of (2.1). The first step consists of a trapezoidal time integration by which the system is  
 756 advanced from  $t_n$  to  $t_{n+\gamma} = t_n + \gamma \Delta t_n$  ( $0 < \gamma < 1$ ):

$$C^{n+\gamma} - \gamma \frac{\Delta t_n}{2} F^{n+\gamma} = C^n + \gamma \frac{\Delta t_n}{2} F^n. \quad (4.6)$$

757 The system is then advanced from  $t_{n+\gamma}$  to  $t_{n+1}$  using the second order differentiation (BDF2):

$$C^{n+1} - \frac{1-\gamma}{2-\gamma} \Delta t_n F^{n+1} = \frac{1}{\gamma(2-\gamma)} C^{n+\gamma} - \frac{(1-\gamma)^2}{\gamma(2-\gamma)} C^n. \quad (4.7)$$

758 In equations (4.6) and (4.7) we have used the notation  $C^{n+\gamma} = C(x, t_n + \gamma \Delta t_n)$  and  $C^{n+1} = C(x, t_n +$   
 759  $\Delta t_n)$ , for the discretization of time.

760 The derivation of the weak form for the coupled system is shown in Appendix C. For the  
 761 TR step, the system is advanced by

$$\int_{\Omega} u_c^{n+\gamma} v_c d\Omega - \gamma \frac{\Delta t_n}{2} (A_{\text{NP}}^{n+\gamma} + A_{\text{P}}^{n+\gamma}) = \int_{\Omega} u_c^n v_c d\Omega + \gamma \frac{\Delta t_n}{2} (A_{\text{NP}}^n + A_{\text{P}}^n), \quad (4.8)$$

762 and, for the BDF2 time step we get

$$\int_{\Omega} u_c^{n+1} v_c d\Omega - \frac{1-\gamma}{2-\gamma} \Delta t_n (A_{\text{NP}}^{n+1} + A_{\text{P}}^{n+1}) = \frac{1}{\gamma(2-\gamma)} \int_{\Omega} u_c^{n+\gamma} v_c d\Omega - \frac{(1-\gamma)^2}{\gamma(2-\gamma)} \int_{\Omega} u_c^n v_c d\Omega. \quad (4.9)$$

763 In Eqs. (4.8) and (4.9),  $u_c$  and  $u_p$  are trial functions for  $C$  and  $\phi$ , respectively,  $v_c$ ,  $v_p$  are  
 764 test functions for  $C$  and  $\phi$ , respectively, and  $A_{\text{NP}}$  and  $A_{\text{P}}$  are the linear forms corresponding to the  
 765 the weak form of  $F[C(t), \phi(t), t]$  in Eq. (4.5), and the weak formulation of Poisson's equation,

766 respectively (See Appendix B):

$$A_{\text{NP}} = -D \int_{\Omega} \nabla u_c \cdot \nabla v_c d\Omega + D \int_{\partial\Omega} (\nabla u_c \cdot \hat{\mathbf{n}}) v_c ds - \mu \int_{\Omega} u_c \nabla u_p \cdot \nabla v_c d\Omega + \mu \int_{\partial\Omega} (\nabla u_p \cdot \hat{\mathbf{n}}) u_c v_c ds \quad (4.10a)$$

$$A_{\text{P}} = - \int_{\Omega} (\nabla u_p \cdot \nabla v_p) d\Omega + \int_{\partial\Omega} (\nabla u_p \cdot \hat{\mathbf{n}}) v_p ds + \frac{q}{\epsilon_{\text{ins}}} \int_{\Omega} u_c v_p d\Omega \quad (4.10b)$$

767

### 768 **4.3 Modeling charge transport in SiO<sub>2</sub>**

769 To validate the model, we ran transport simulations in SiO<sub>2</sub>, to reproduce the experimental  
 770 trends described by Snow et al. [69]. An initial distribution of Na<sup>+</sup> with a surface concentration  
 771  $Q_0 = 29 \times 10^{11} / \text{cm}^{-2}$ , uniformly distributed over the first 140 Å of a 0.2 μm-thick layer of SiO<sub>2</sub>  
 772 was assumed. The diffusion coefficient of Na<sup>+</sup> was set to  $5 \times 10^{-16} \text{ cm}^2/\text{s}$  and no-flux boundary  
 773 conditions for the concentration were set to

$$-D\nabla C - \mu C \nabla \phi = 0 \quad (4.11)$$

774 at  $x = 0$  and  $x = L = 0.2 \mu\text{m}$ . The gradient of the potential  $\nabla \phi$  is obtained self-consistently from  
 775 (4.4). For the electric potential we set  $\phi(x = 0) = 10 \text{ V}$  and  $\phi(x = L) = 0$ . As an initial condition,  
 776 we set a uniform finite concentration of ions in the layer to a negligible value (we used  $C \rightarrow$   
 777  $10^{-20} / \text{cm}^3$ ) and a linear potential drop within the dielectric, consistent with a potential at the gate  
 778 of  $\phi(x = 0) = 10 \text{ V}$ .

779 Equations (4.8) and (4.9), together with the boundary conditions are solved self-consis-  
 780 tently. At each time step,  $\nabla \phi$  was estimated from (4.4), by integrating the latest concentration  
 781 profile. The mesh was refined closed to the boundaries with a minimum element size  $\Delta x = 1.67 \text{ \AA}$

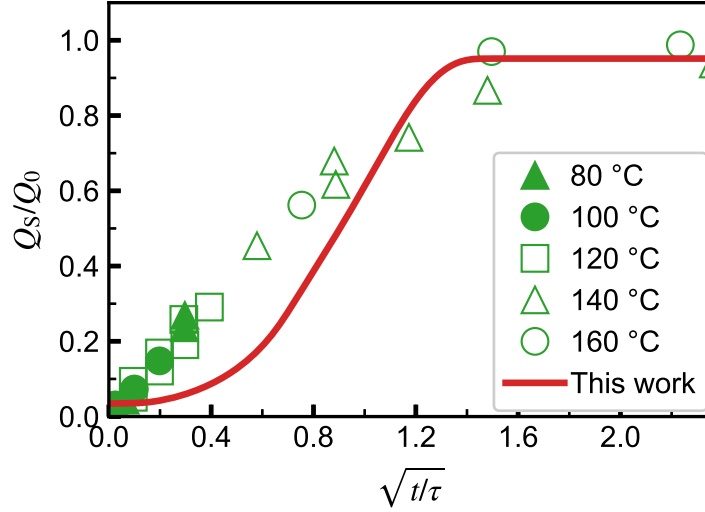


Figure 4.1: Simulated image charge in the semiconductor of a MIS structure as a function of time. Experimental points are taken from Snow et al. [69].

782 and a maximum element size  $\Delta x = 6.67 \text{ \AA}$ . The time step was set to a fixed value of  $\Delta t = 50 \text{ s}$  to  
 783 reduce time truncation error in the TR-BDF2 integration as much as possible. The solution was  
 784 computed using FEniCS [82, 88] and the auxiliary numerical tools for this framework [89–94].  
 785 The results for a  $\sim 6.25 \text{ h}$  drift-diffusion simulation at a  $140 \text{ }^\circ\text{C}$  temperature stress are shown in  
 786 Fig. 4.1. The normalized image charge in the semiconductor as a function of dimensionless time  
 787  $t/\tau$ , predicted by the model is plotted in red, together with the experimental excess surface charge  
 788 in the semiconductor, reported by Snow et al. [69]. The model successfully predicts a linear rela-  
 789 tionship of  $Q_S/Q_0$  as a function of  $\bar{t} = \sqrt{t/\tau}$  for  $\bar{t} \lesssim 1.4$ , after which the curvature changes and  
 790 the surface concentration saturates. This is consistent with the boundary layer approximation [69].  
 791 From the boundary layer theory [69]

$$\left| \frac{Q_S}{Q_0} \right| \approx \frac{4}{\pi^{3/2}} \left( \frac{t}{\tau} \right)^{1/2}, \text{ for } t \ll \tau \quad (4.12)$$

$$\left| \frac{Q_S}{Q_0} \right| \approx 1 - \frac{8}{\pi^2} \exp\left(-\frac{t}{\tau}\right), \text{ for } t \gg \tau, \quad (4.13)$$

792 with

$$\tau \equiv 4x_1^2/\pi^2 D.$$

793 The value of  $x_1$  in Eq. (4.14) corresponds to the thickness of the boundary layer. Eq. (4.12) es-  
794 tablishes the slope and time constant for the linear and exponential regime. Deviation from the  
795 experimental slope arises from a different constant of proportionality originating from (1) the dis-  
796 crepancy introduced by the boundary layer width  $x_1$  which is no longer a constant in the numerical  
797 simulation and, (2) on the effective magnitude of the electric field when taking the superposition  
798 of the applied electric stress and the electric potential due to the ion concentration.

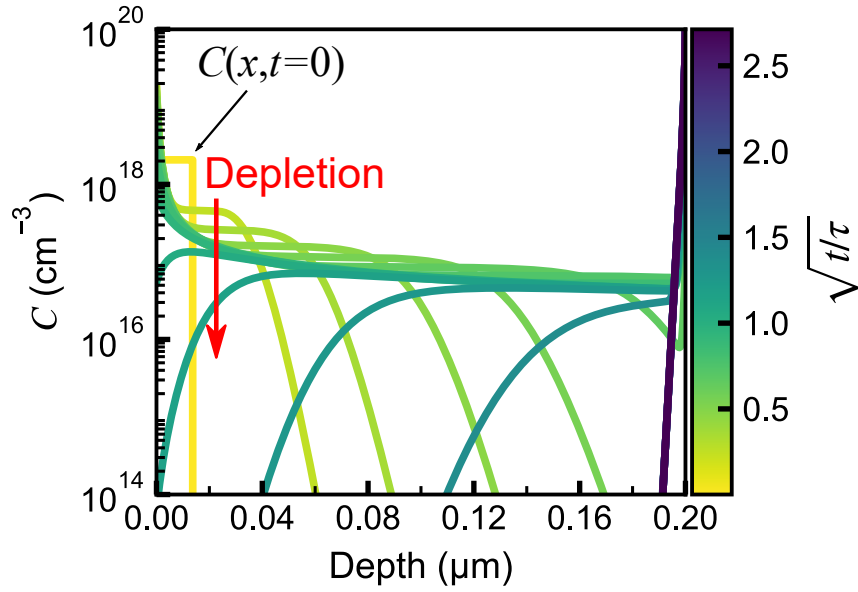


Figure 4.2: Selected concentration profiles corresponding to the image charge plotted in Fig. 4.1.

799

In Fig. 4.2, a selection of concentration profiles used to compute the image charge in

800

Fig. 4.1 is plotted as a function of time indicated in color scale. Shown in yellow is the shape

801

of the initial concentration profile. At early times, we notice that the initial profile is pulled in

802

opposite directions: the image charge at the gate pulls in the negative direction of  $x$ , and the applied

803

bias drifts the ions in the positive direction. Because of the balancing effect of the image charge

804

potential and the applied bias, the region closest to the gate is in fact dominated by diffusion. As a

805

consequence of the image charge effect drifting  $\text{Na}^+$  driving ions towards the gate, and the applied

806

bias driving ions towards the Si interface, the concentration around the original profile depletes.

807

Then, the magnitude of  $\mu EC$  decreases, leading to slower transit times during the early stages of

808

the simulation, as is evidenced in the region for  $t \lesssim 0.25\text{h}$  in Fig. 4.1. At intermediate times,

809

$\text{Na}^+$  piles up at the semiconductor interface driven by the applied bias and the local image charge

810

effect at that interface. Ions are lost from the region around the vicinity of the gate to the drift



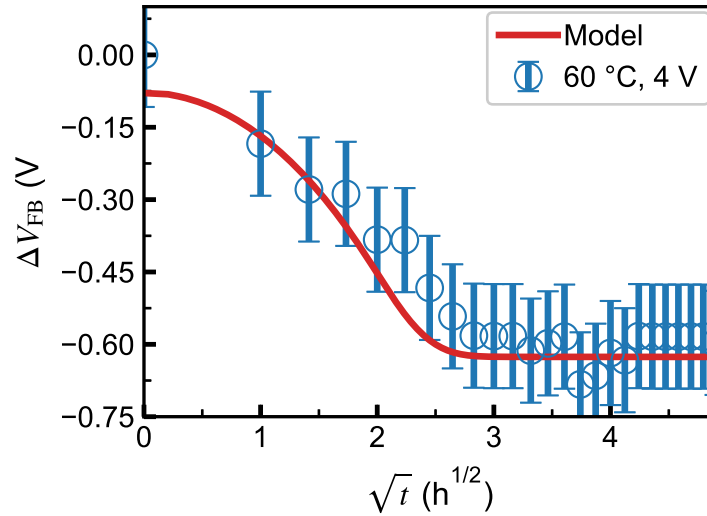


Figure 4.3: Simulated  $V_{FB}$  as a function of time for a  $\text{SiO}_2$  MIS structure using  $D = 4 \times 10^{-17} \text{ cm}^2/\text{s}$ . Empty circles indicate the experimental results of samples stressed at  $60^\circ\text{C}$  under a bias of 4 V.

811 dominated region, driven by a large concentration gradient close to this interface. This causes  
 812 further depletion of the gate interface, which is identified by profiles with positive concentration  
 813 gradients moving towards the semiconductor interface. This result quantitatively describes the  
 814 kinetics of Na migration in MIS structures, suggesting that the initial concentration profile might  
 815 lay further into the bulk of  $\text{SiO}_2$  (leading to lower initial concentration which in turn reduces image  
 816 charge effect at the gate).

817 To provide further validation of the model with more recent experimental data, following  
 818 the experiment by Snow et al. [69], we prepared  $\text{SiO}_2$  MIS capacitors by dry oxidation of n-type  
 819 Si (100) wafers at a temperature of  $1100^\circ\text{C}$ . Prior to oxidation, the silicon wafers were cleaned  
 820 with acetone, ethanol and isopropyl alcohol, followed by RCAI and RCAII cleaning procedures.  
 821 The thickness of the  $\text{SiO}_2$  layer was mapped by spectroscopic ellipsometry. An average thickness  
 822 of 100 nm was measured. Sodium contamination was introduced by thermal evaporation at  $800^\circ\text{C}$

823 from a NaCl source, followed by contact deposition without breaking the vacuum. The rate of  
824 deposition of the NaCl layer was monitored using a quartz microbalance up to value of 25 nm.  
825 A pattern of  $\sim 1$  mm diameter circles with a pitch of 2.54 mm was deposited using a shadow  
826 mask. Approximately 200 nm of aluminum were deposited as front and back contacts. The bottom  
827 surface of the wafer was swabbed with hydrofluoric acid prior to metallization.

828 The as-prepared samples were subject to a bias-temperature stress of 60 °C and 4 V in  
829 intervals of 1 h. Between stress intervals, the samples were cooled down to room temperature and  
830 capacitance-voltage measurements were collected. The flatband voltage shift  $\Delta V_{\text{FB}}$  was extracted  
831 as described in our previous work [62].

832 To simulate the transport of  $\text{Na}^+$  in the  $\text{SiO}_2$  MIS capacitor, we used a  $Q_0 = 1.4 \times 10^{11} \text{ cm}^{-2}$ .  
833 This value was selected based on a quick estimation of the expected saturation  $V_{\text{FB}} \approx -0.6 \text{ V} \approx$   
834  $-Q_0/\epsilon_{\text{ins}}L$  from the experimental results. A diffusivity of  $4 \times 10^{-17} \text{ cm}^2/\text{s}$  was used to match the  
835 temporal dependence of  $\Delta V_{\text{FB}}$ . For comparison, this value is two orders of magnitude larger than  
836 the one we would expect from the extrapolation of the Arrhenius relationship reported by Frischat  
837 [95] taken within the range of 170 °C to 250 °C for commercial  $\text{SiO}_2$  glasses. The initial concentra-  
838 tion profile was set to reach a 25 nm depth from the gate, in accordance to the monitored thickness  
839 during NaCl evaporation. As can be seen in Fig. 4.3, the simulated profile matches well the exper-  
840 imental data within errorbars. For consistency with Fig. 4.1, the results were plotted against  $t^{1/2}$ .  
841 In this representation, a quadratic dependence of  $\Delta V_{\text{FB}}$  is indicative of drift-dominated transport.

## 4.4 Behavior before the characteristic transit time

In section 3.6, the concept of characteristic transit time was introduced to account for a simplified behavior whenever the concentration was below the threshold value  $C_{\text{th}} \equiv 2\epsilon_{\text{ins}}/qL^2V_{\text{stress}}$ . To cross reference our transport model with this behavior at short times and under low concentrations we modified the boundary conditions to accommodate for a constant source of ions at the gate. Then, we ran simulations on a MIS structure with a thickness of 80 nm at a temperature of 80 °C, with an applied bias of 4 V. The corresponding concentration profiles as a function of time are shown in Fig. 4.4, where the drift-diffusion time is indicated in color scale. Fig. 4.4(a) illustrates the transport kinetics at short times as modeled by the numerical solution to Poisson-Nernst-Planck coupled system with a constant source and a closed boundary at the semiconductor interface. The estimated characteristic time for this system is  $\tau_c \approx 8.4$  h, beyond which the concentration profile begins to pile up at the semiconductor interface, due to the effect of drift and accumulation of negative image charges in the semiconductor. As expected from the boundary conditions, we do not see depletion at the metal gate interface, in contrast with the closed system described in Section 4.3. In fact, when  $t < \tau_c$ , this boundary condition behaves identically to the approximation at short times given by (3.7), shown in Fig. 4.4(b). A variation of only 2.8 mV in  $V_{\text{FB}}$  at  $t = \tau_c$  was estimated between the two models, which is less than the thermal voltage  $k_{\text{B}}T \approx 30$  mV at the simulated temperature. This is expected due to the fact that, for the given bias and thickness, our threshold concentration is  $C_{\text{th}} \approx 5 \times 10^{17} \text{ cm}^{-3}$ , compared to our average concentration of  $10^{16} \text{ cm}^{-3}$ .

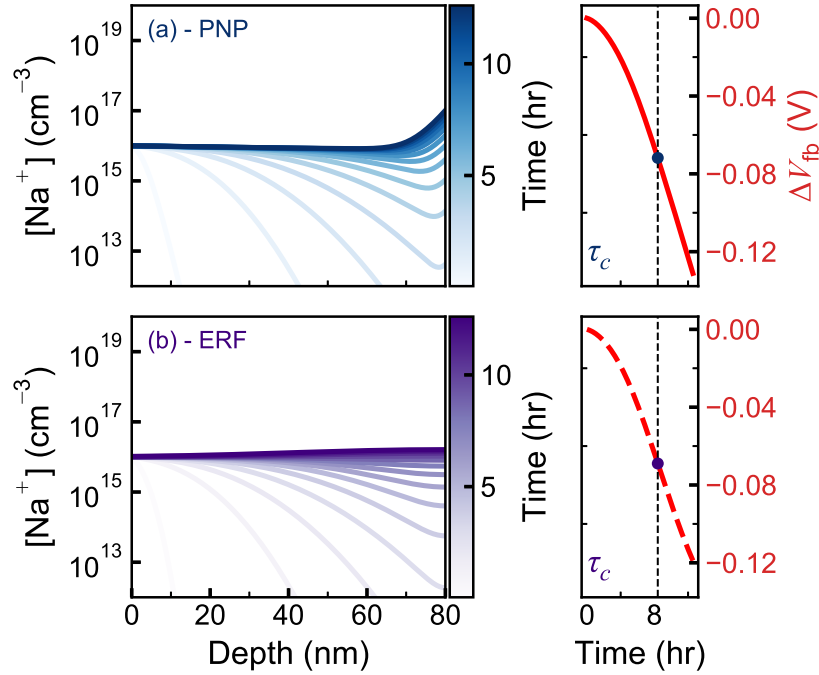


Figure 4.4: Concentration profiles as a function of time, indicated in color scale, modeled by (a) a FEM simulation of the full Poisson-Nernst-Planck coupled system (PNP) and (b) the approximate solution (ERF) at short times given by (3.7). To the right of each pane is plotted the corresponding  $\Delta V_{\text{FB}}$  due to the ionic charge distribution as a function of time. In the same plot is indicated the characteristic transit time within the approximation described in Sec. 3.6. The difference between the estimated  $V_{\text{FB}}$  at  $t = \tau_c$  is of only 2.8 mV.

## 862 4.5 Na transport in stacks of materials

863 The finite element implementation of the present model can be easily re-formulated to study  
 864 the kinetics of transport of ionic species in stacks of materials. We can model the transport of  $\text{Na}^+$   
 865 through the dielectric and into the silicon, by treating both systems independently, and connecting  
 866 them with a flux boundary condition. Given the discontinuity in the diffusion coefficients, impurity  
 867 equilibrium concentrations and electrical properties in the layer stack, segregation needs to be  
 868 considered. To account for impurity segregation at the dielectric/semiconductor interface we use a

869 phenomenological, first order kinetics description of the flux [96]:

$$J_s \equiv h \left( C_{\text{ins}}(L, t) - \frac{C_{\text{Si}}(L, t)}{m} \right), \quad (4.14)$$

870 where  $L$  is the thickness of the insulating film,  $C_{\text{ins}}(L, t)$ ,  $C_{\text{Si}}(L, t)$  are the impurity concentrations  
 871 defined in the insulator and in the semiconductor, respectively,  $h$  is the surface mass transfer coef-  
 872 ficient given in units of distance over unit time, and

$$m \equiv \left( \frac{C_{\text{Si}}}{C_{\text{ins}}} \right) \text{ in equilibrium} \quad (4.15)$$

873 is the segregation coefficient.

874 The total flux at  $x = L$  from the dielectric side is given by

$$-D_{\text{SiN}} \nabla C - \mu C \nabla \phi = \mathbf{J}_s. \quad (4.16)$$

875 Because silicon is assumed to have an electrical resistivity several orders of magnitude  
 876 less than the insulating films (from  $\sim 10^{-3} \Omega \text{ cm}$  to  $\sim 10^5 \Omega \text{ cm}$ , compared to typical insulating  
 877 resistivities on the order of  $10^{11} \Omega \text{ cm}$ ), it is possible to neglect drift in silicon. As it turns out, in  
 878 PV devices, the high doping concentration of the emitter side puts an upper bound on the resistivity  
 879 of Si on the order of  $10^{-2} \Omega \text{ cm}$ . Hence, transport in silicon is governed by pure diffusion. And  
 880 the total flux at  $x = L$  from the side of silicon is given by

$$-D_{\text{Si}} \nabla C = \mathbf{J}_s. \quad (4.17)$$

881 Eqs. (4.16) and (4.17) are introduced easily as Neumann boundary conditions on the surface  
 882 element terms  $ds$  in Eq. (4.10).

883 Incoming flux at the metal gate interface can also be easily instrumented in the model by  
 884 considering a source of ions with surface concentration  $S$  given in atoms per unit area flowing in

885 to the insulating layer at a rate  $k$  in units of inverse time:

$$J_{\text{source}} = kS \quad (4.18)$$

886 To account for the finite size of the source, we consider that the source is depleted after a time  $1/k$   
887 after which we set  $J_{\text{source}} = 0$ .

888 To demonstrate transport through stacks of materials, we simulated the kinetics of  $\text{Na}^+$   
889 through the  $\text{SiN}_x/\text{Si}$  stack by means of Eqs. (4.8) and (4.9), and setting the boundary conditions  
890 defined by equations (4.16), (4.17) and (4.18). The thickness of the  $\text{SiN}_x$  was set to 75 nm, in  
891 accordance to typical values used in PV modules [27]; and the thickness of silicon was set to 1  $\mu\text{m}$   
892 which is long enough to be considered infinite within the timescale of the simulation (96 h) for  
893 the selected value of the diffusivity of Na in Si. We set the temperature to 85 °C and considered  
894 a diffusivity of  $\text{Na}^+$  in  $\text{SiN}_x$  of  $\sim 4 \times 10^{-16} \text{cm}^2/\text{s}$  in accordance to our trapping-corrected BTS  
895 estimates (Gastrow et al. [62]), and a value of  $10^{-16} \text{cm}^2/\text{s}$  for the diffusivity of Na in Si. The latter  
896 is larger than the expected values of  $\sim 10^{-21} \text{cm}^2/\text{s}$  from extrapolation of experimental Arrhenius  
897 curves taken at temperatures above 1000 °C [97], but lower than the bounds estimated based upon  
898 typical failure times of PID-s affected PV modules, which put an upper bound of  $\sim 10^{-7} \text{cm}^2/\text{s}$   
899 [98]. An electric stress of 75 mV was applied, corresponding to an electric field of  $10^4 \text{V cm}$ ,  
900 which is in range with the estimations from the voltage divider model shown in Table 2.1, for  
901 low resistivity nitrides. For the source term, we chose a surface concentration of  $S = 10^{10} \text{cm}^{-2}$   
902 which is two orders of magnitude less than the values corresponding to intentional contamination  
903 in [32, 69], and a rate of ingress  $k = 10^{-5} \text{s}^{-1}$ , for which the source is partially depleted within the  
904 simulation time of  $\sim 3.5 \times 10^5 \text{s}$  (96 h). For the segregation flux, we used a surface mass transfer  
905 coefficient of  $h = 10^{-12} \text{cm s}^{-1}$  and a segregation coefficient  $m = 1$ . The results are plotted in

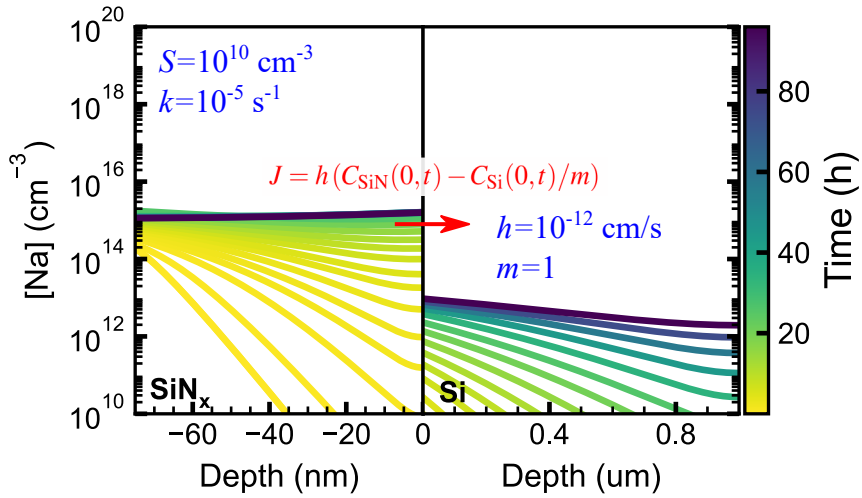


Figure 4.5: Simulated concentration profiles as a function of time across a SiN<sub>x</sub>/Si material stack.

906 Fig. 4.5, where the concentration profiles are shown as a function of time in color scale. Notice  
 907 that, because the flux at the source is maintained constant and the rate is  $k = 10^{-5} \text{ s}^{-1}$ , depletion  
 908 becomes evident after the  $\sim 28 \text{ h}$ . In fact, the concentration profile almost flattens within the SiN<sub>x</sub>  
 909 over the course of the simulation. Furthermore, because the diffusion coefficient in the silicon  
 910 is set to a higher value ( $10^{-14} \text{ cm}^2/\text{s}$ ) with respect to the value in SiN<sub>x</sub> ( $4 \times 10^{-16} \text{ cm}^2/\text{s}$ ), Na<sup>+</sup>  
 911 is transferred to silicon and is diffused into the bulk of Si relatively easily. Additionally, due to  
 912 the low magnitude of the electric field in SiN<sub>x</sub>, no accumulation of Na is observed at the SiN<sub>x</sub>/Si  
 913 interface, and ionic transport in the dielectric film is dominated by diffusion, as can be observed in  
 914 Fig. 4.5, by the slopes of the concentration profiles.

## 915 **4.6 Conclusions**

916 We developed a model to simulate the kinetics of ion diffusion in MIS devices. Validation  
917 with experimental results resulted in good agreement of the simulated trend of the image charge  
918 in the semiconductor interface. Reference with BTS experiments also results in agreement of the  
919 predicted  $\Delta V_{FB}$  due to ion migration. Under the limiting case of low concentration and short times,  
920 we verified that the solution can be correctly approximated using a constant electric field which  
921 simplifies to a pure diffusion problem, as described in Sec. 3.3. The estimated solution to Poisson-  
922 Nernst-Planck coupled system yields all the information required to study ion transport kinetics  
923 with respect to the the charge distribution. This is especially important to visually understand  
924 the assumptions behind boundary layer theory and extend upon it. Our model allows for direct  
925 extrapolation to other type of dielectric structures and material stacks which are of interest for  
926 modeling reliability of semiconductor devices.

927 Chapter 4, in part, is being prepared for submission in the *Journal of Applied Physics*  
928 under the title “Numerical Solution to Poisson-Nernst-Planck Equations for Ion Transport in MIS  
929 structures and Solar Cells” by E. Martinez-Loran, G. von Gastrow, J. Clenney, F. Contreras-Torres,  
930 R. Meier, M.I. Berton, P. Bandaru and D.P. Fenning. The dissertation/thesis author was the primary  
931 investigator and author of this paper.



## 932 **5 | Device degradation modeling**

933 The photovoltaic industry has made huge efforts to engineer efficient and reliable solar  
934 modules and to test the long-term performance of modules. Reliability tests have shown potential-  
935 induced degradation can lead device failure before the average warranty time in silicon solar mod-  
936 ules [1, 3–5]. Over the last decade, much work has been put to standardize reliability tests [5, 7,  
937 14, 24, 27, 68, 99–102] and to develop predictive models to estimate the progression of PID [14,  
938 15, 68].

### 939 **5.1 Introduction**

940 Power law models have demonstrated some use to describe the loss in power output, [21,  
941 36, 38, 68] albeit with limitations on the applicability to weeks-long simulations. Since PID in p-Si  
942 solar modules has been ascribed to electrical shunting of the emitter, empirical models describing  
943 the progression of the shunt resistance  $R_{sh}$  have phenomenologically described the progression of  
944  $R_{sh}$  with stress time, and even incorporated environmental conditions [14, 29, 103]. More recently  
945 Schulze et al. [15], proposed a simplified drift-diffusion model of a shunt that adequately fits  
946 the PID progression and recovery in  $R_{sh}$  for p-Si modules. While this model provides insights  
947 into the physical parameter space that impacts ion transport and can potentially be engineered to

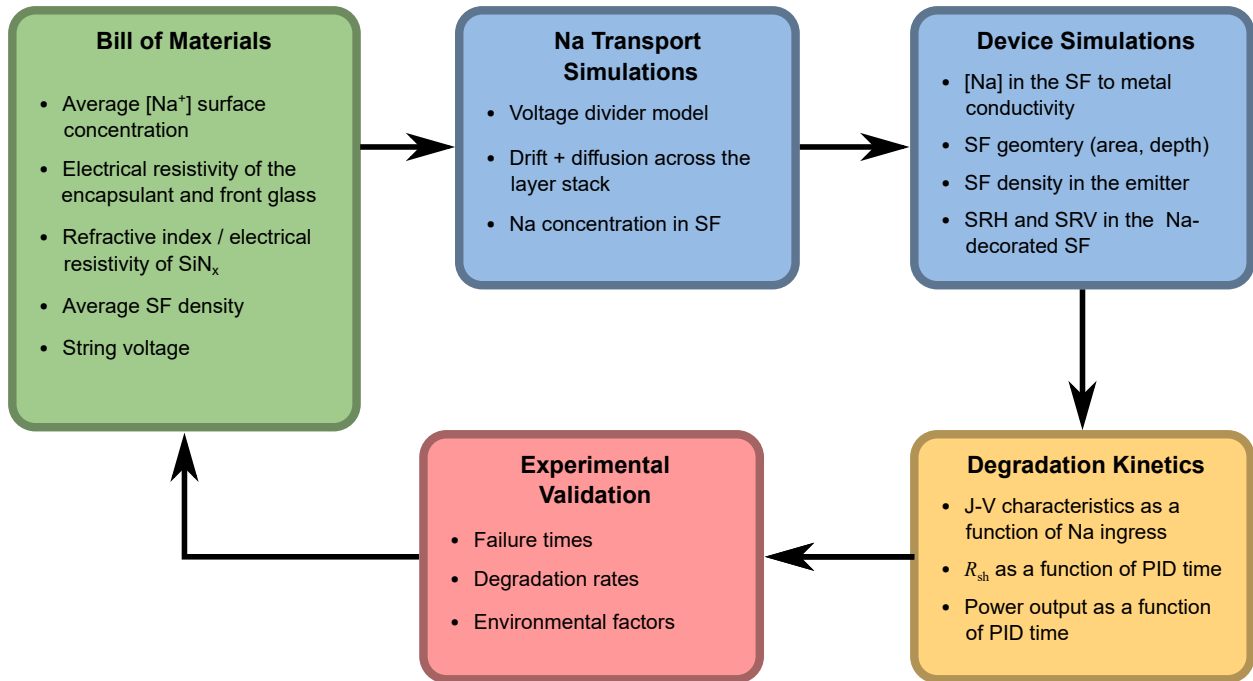


Figure 5.1: Proposed modeling framework aimed at predicting PID-s and engineering PID-robust PV modules.

948 reduce PID, the introduced level of simplification weakens the connection to describe engineerable  
 949 parameters, which can be determined through the bill of materials.

950 The assumed mechanism behind PID-s in p-Si PV modules is that Na contamination from  
 951 different depths in the module migrates to the silicon emitter where it is responsible for shunting  
 952 behavior [5, 9, 12, 13, 18, 19]. Microscopy analyses have established that Na-decorated {111}  
 953 stacking faults in p-type silicon lead to the formation of electrical shunts in the emitter of the solar  
 954 cell leading to PID [12, 19]. It has been proposed that the introduction of Na in these defects occurs  
 955 through electric-field assisted transport, whereupon Na contamination in the silicon nitride (SiN<sub>x</sub>)  
 956 anti-reflective coating of the solar cell moves towards the emitter of the solar cell [12]. Once Na  
 957 reaches silicon, it tends to concentrate in the {111} stacking faults [9, 12, 13, 19]. These defects  
 958 shunt the p-n junction and result in the loss of carrier collection efficiency [9, 12, 19, 104].

959 In this chapter we describe a mechanistic model to describe the formation of the electrical

960 shunts in the module which takes in to account the full kinetics of ion transport formulated in  
961 Chapter 4. The present model allows to parametrize the power output degradation and  $R_{sh}$  in terms  
962 of intrinsic properties, like the diffusion coefficients of Na in the different layers, and engineerable  
963 parameters like the resistivity of the encapsulant layer and the incorporation of blocking layers  
964 between the  $SiN_x$  and the Si emitter.

965         The overarching goal of this modeling framework is to develop a full physical analysis  
966 aimed at understanding the impact of the properties of the materials in the PV module, the operating  
967 voltages, and environmental conditions in PID-s. As proposed in Fig. 5.1, this iterative analysis  
968 allows a comprehensive study of the multiple factors leading to PID-s. This methodology will  
969 provide guidance to optimize the bill of materials for PID-robust PV modules. The proposed inputs  
970 of the model are the electrical resistivity of the front-cover glass, encapsulant layer and  $SiN_x$ . As  
971 a proxy for the electrical resistivity of  $SiN_x$ , refractive index measurements have shown good  
972 correlation with PID-s resistance [5, 27, 59, 74, 105, 106]. Also of interest, is the average surface  
973 concentration of stacking fault defects in the emitter of the module and, the string voltage, which  
974 determines to what extent ions drift towards the emitter. These parameters can be used in transport  
975 simulations to model the ingress of Na through the module stack and determine the concentration  
976 profile of Na in the stacking faults. The Na concentration in the defect and this information,  
977 together with the area of the shunt and density of defects is mapped to a shunt conductivity and  
978 the kinetics model is used by a device simulation tool that determines the current-voltage (J-V)  
979 characteristic of the PV module. The kinetics of degradation is then estimated by analyzing the  
980 J-V characteristic of the device. Simulated  $R_{sh}$  and power output is subsequently validated against  
981 experimental data.

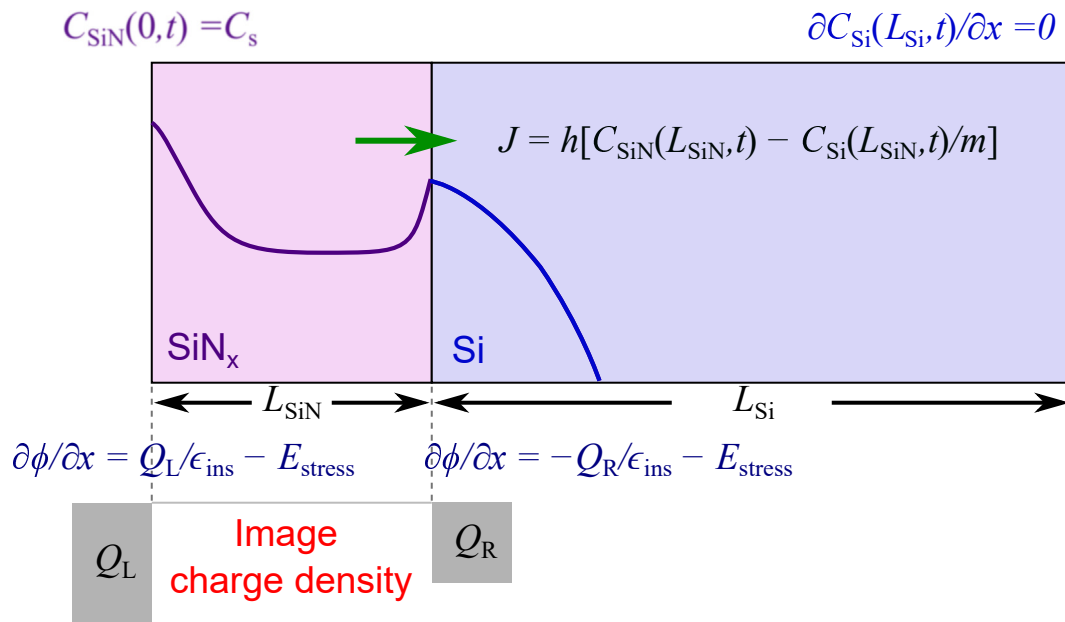


Figure 5.2: Boundary conditions used for Na transport simulations in p-Si PV modules. As a first approximation we set a constant source boundary condition at the EVA/SiN<sub>x</sub> located at  $x = 0$ . At the bottom of Si, located at  $x = L_{\text{Si}}$ , a zero-flux boundary condition is imposed. The two materials are connected by a first order kinetics segregation flux boundary condition. The boundary condition for the electric potential is determined by the electric field induced by the image charges at the EVA and Si interfaces, due to the Na<sup>+</sup> concentration in SiN<sub>x</sub>.

## 5.2 Modeling Na migration kinetics

In PID-s, Na contamination is assumed to reach the  $n^+$ -Si emitter of the PV module, where it “decorates” the stacking faults, after migrating through the encapsulation and the  $\text{SiN}_x$  anti-reflective coating. The process is driven by the large potential difference between the cell at a floating negative potential  $V \geq 1000\text{ V}$  and the metal frame which is grounded at  $V = 0$  [5, 12].

We assume that the electric field in the module stack can be estimated on the basis of the voltage divider introduced in Sec. 1.4. In Table 2.1, we show typical ranges of the equivalent electric field in each material of the PV module, based on typical values of the thickness.

The origin of Na contamination in solar modules has been historically ascribed to the front cover soda-lime glass [4, 5, 7]. For instance, it has been shown that replacement of soda-lime glass with a quartz glass window mitigates pseudo fill factor losses in Si modules subject to damp heat-PID tests [7, 18]. Nevertheless, more recent reports show that PID can occur in the absence of a front-cover glass and even in the absence of an encapsulant [23–25]. This suggests that Na contamination might have been introduced during lamination of the module. This hypothesis has been supported by the presence of Na contamination within EVA, particularly, close to the EVA/ $\text{SiN}_x$  interface, prior to PID testing [26]. In Sec. 2.2 we provided further evidence that, migration through the EVA might have a limited impact on PID-s due to the fact that Na does not drift through the whole thickness of the encapsulant in the time scale of PID degradation. Therefore, we assume that Na contamination is present at the EVA/ $\text{SiN}_x$  prior to PID stressing.

It is assumed that PID-s occurs by Na contamination present at the EVA/ $\text{SiN}_x$  interface being drifted through the  $\text{SiN}_x$  anti-reflective coating and diffused into the  $n^+$ -Si emitter of the PV

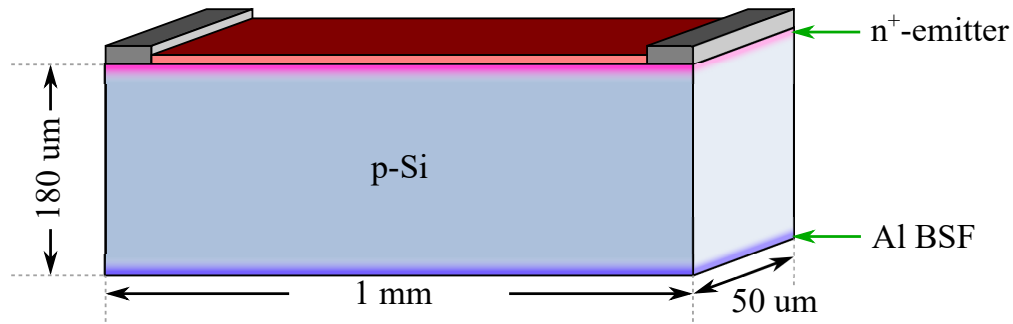


Figure 5.3: Schematic of the Aluminum Back Surface used to simulate the module degradation.

1003 module. Transport in  $\text{SiN}_x$  is governed by drift and diffusive flux. The relative contribution of the  
 1004 drift and diffusive terms is determined by the magnitude of the electric field in the  $\text{SiN}_x$ .

1005 Transport is then simulated using the FEM modeling framework described in Sec. 4.5.  
 1006 The boundary conditions that we used are described in Fig. 5.2. For small source concentrations,  
 1007 we assume that a constant source boundary condition can suffice to describe ion ingress at the  
 1008 EVA/ $\text{SiN}_x$  interface. At the bottom of the Si simulation volume, we impose a zero-flux boundary  
 1009 condition. Transport is modeled independently in each layer with a coupling boundary condition  
 1010 given by a first order kinetics segregation flux [96]. The electric field is assumed to be negligible  
 1011 in the  $n^+$ -Si emitter, hence transport is well described by pure diffusion in this material. In  $\text{SiN}_x$ ,  
 1012 the electric stress is usually non-negligible and transport is governed by Nernst-Planck's equation.  
 1013 Due to the introduction of  $\text{Na}^+$ , the electric potential is determined by the superposition of the  
 1014 applied bias and the potential induced by the  $\text{Na}^+$  charge distribution. Therefore, as discussed in  
 1015 Sec. 3.6, the complete solution for times  $t > \tau_c$  is coupled to the solution of Poisson's equation for  
 1016 the charge distribution in  $\text{SiN}_x$ . We set the Neumann boundary conditions to the electric potential  
 1017 self-consistently, based on the determination of the image charges at the EVA and Si interface due  
 1018 to the  $\text{Na}^+$  concentration profile.

## 1019 **5.3 Device simulations**

1020 To determine the amount of degradation as a function of Na ingress into the silicon sub-  
1021 strate, we model the effect of introducing an electrical shunt in the emitter of a typical p-Si, alu-  
1022 minum back surface field (Al-BSF) PV module. The schematic of the simulated devices is shown  
1023 in Fig. 5.3. The boron concentration in the base, emitter phosphorus concentration and Al-BSF  
1024 concentration used in the simulation are listed in Table 5.1. Both, the emitter and Al-BSF are  
1025 defined using a Gaussian concentration profile which reaches a value of  $10^{16} \text{ cm}^{-3}$  at a depth of  
1026 300 nm from the front and back surfaces, respectively. Device simulations are performed using  
1027 Synopsys Sentaurus FEM solver<sup>1</sup>. Carrier generation is estimated using OPAL<sup>2</sup> at 1 Sun, consid-  
1028 ering an  $\text{SiN}_x$  anti-reflective coating of 75 nm. Shockley-Read-Hall bulk recombination is activated  
1029 and a bulk lifetime of 5 ms. The selected bulk lifetime is consistent with several reports of high  
1030 quality float zone Si wafers at the same level of doping [107]. The surface recombination velocity  
1031 at the  $\text{SiN}_x/\text{Si}$  interface was estimated using Altermatt's parametrization [108], and Fermi-Dirac  
1032 statistics is activated to account for the large doping concentration at the emitter.

1033 To model PID-s, we introduce a metallic structure within the emitter of the cell as illustrated  
1034 in Fig. 5.4. We define the conductivity profile along the SF according to the spreading resistance  
1035 measurements taken by Korol [97] on Na-implanted silicon. To avoid introducing spurious shunt-  
1036 ing at early stages of the simulation due to the presence of a metallic structure, we neglected the  
1037 regions of the shunt with conductivity less than  $10^{-10} \text{ S cm}^{-1}$ . Concentration profiles as a function  
1038 of time were modeled according to the parameters in Table 5.1.

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<sup>1</sup><http://www.synopsys.com/home.aspx>

<sup>2</sup>[https://www2.pvlighthouse.com.au/calculators/OPAL\\_2/OPAL\\_2.aspx](https://www2.pvlighthouse.com.au/calculators/OPAL_2/OPAL_2.aspx)

Table 5.1: Parameters used to model transport across the SiN<sub>x</sub>/Si stack.

Parameter	Symbol	Value	Units
Simulation temperature	$T$	85	°C
Thickness of SiN <sub>x</sub>	$L_{\text{SiN}}$	75	nm
Thickness of Si	$L_{\text{Si}}$	1	μm
Electric field in SiN <sub>x</sub>	$E_{\text{SiN}}$	0.5	MV cm <sup>-1</sup>
Electric field in Si	$E_{\text{Si}}$	0	V cm <sup>-1</sup>
Diffusion coefficient of Na <sup>+</sup> in SiN <sub>x</sub> <sup>a</sup>	$D_{\text{SiN}}$	$4 \times 10^{-16}$	cm <sup>2</sup> /s
Concentration of the source	$C_s$	$10^{16}$	cm <sup>-3</sup>
Doping concentration in the p-Si base	$N_A$	$10^{16}$	cm <sup>-3</sup>
Doping concentration in the n-Si emitter	$N_D$	$10^{19}$	cm <sup>-3</sup>
Al-BSF doping level	$N_D$	$10^{19}$	cm <sup>-3</sup>
Surface mass transfer coefficient at the SiN <sub>x</sub> /Si interface	$h$	$10^{-12}$	cm s <sup>-1</sup>
Segregation coefficient at the SiN <sub>x</sub> /Si interface	$m$	1.0	-

<sup>a</sup> Based on high resistivity SiN<sub>x</sub> measurements Gastrow et al. [62].



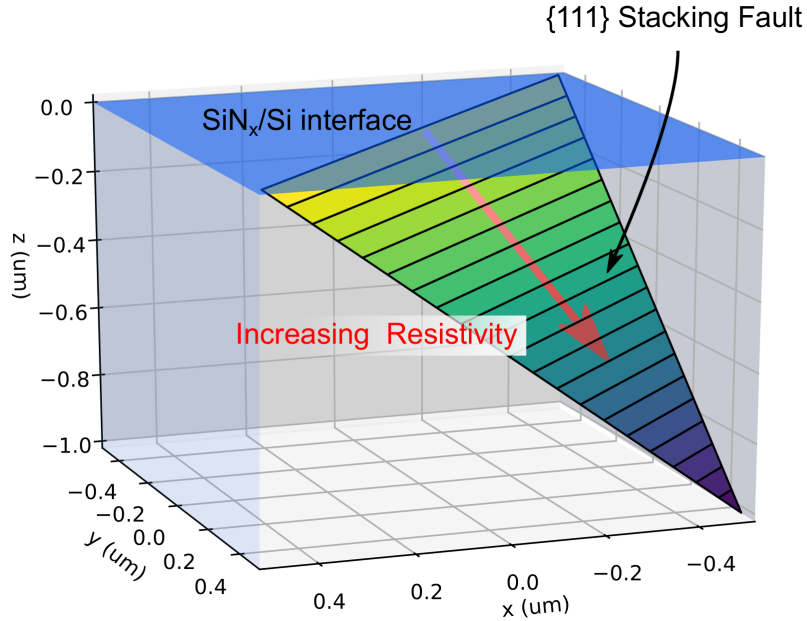


Figure 5.4: Schematic of a metallic shunt introduced in the  $n^+$ -Si emitter of the PV module to model PID-s. The triangular plane represents a metallic region over a  $\{111\}$ -stacking fault, with a resistivity profile determined by the Na concentration.

1039 Fig. 5.5 shows a PID-s kinetics simulation for a device stressed at  $85^\circ\text{C}$  under a voltage  
 1040 equivalent to  $0.5\text{ MV cm}^{-1}$ , according to the parameters described in Table 5.1. Fig. 5.5(a) shows  
 1041 the corresponding concentration profiles in the module stack used in the determination of the shunt  
 1042 conductivity and subsequently, to simulate the J-V characteristic. Due to the small value of the  
 1043 surface mass transfer coefficient ( $h = 10^{-12}\text{ cm s}^{-1}$ ), sinking into the emitter occurs at a slower  
 1044 rate, compared to the rate at which ions drift within the  $\text{SiN}_x$ . This leads to a build-up in the  
 1045 concentration on the  $\text{SiN}_x$  side of the  $\text{SiN}_x/\text{Si}$ , which reaches concentrations up to  $\sim 10^{19}\text{ cm}^{-3}$ .  
 1046 For the selected value of the diffusion coefficient of Na in the SF, Na reaches concentrations  $\gtrsim$   
 1047  $10^{14}\text{ cm}^{-3}$  within a few hours leading to values of the conductivity  $\gtrsim 10^{-3}\text{ S cm}^{-1}$  within the  $p$ - $n$   
 1048 junction of the device, as can be seen in Fig. 5.5(b). The simulated J-V characteristic due to PID  
 1049 is shown in Fig. 5.5(c). The first J-V curve shown in bright yellow, at time zero corresponds to  
 1050 the device in the absence of the shunt. Below the threshold value of  $\sigma_{\text{shunt}} < 10^{-3}\text{ S cm}^{-1}$ , only a

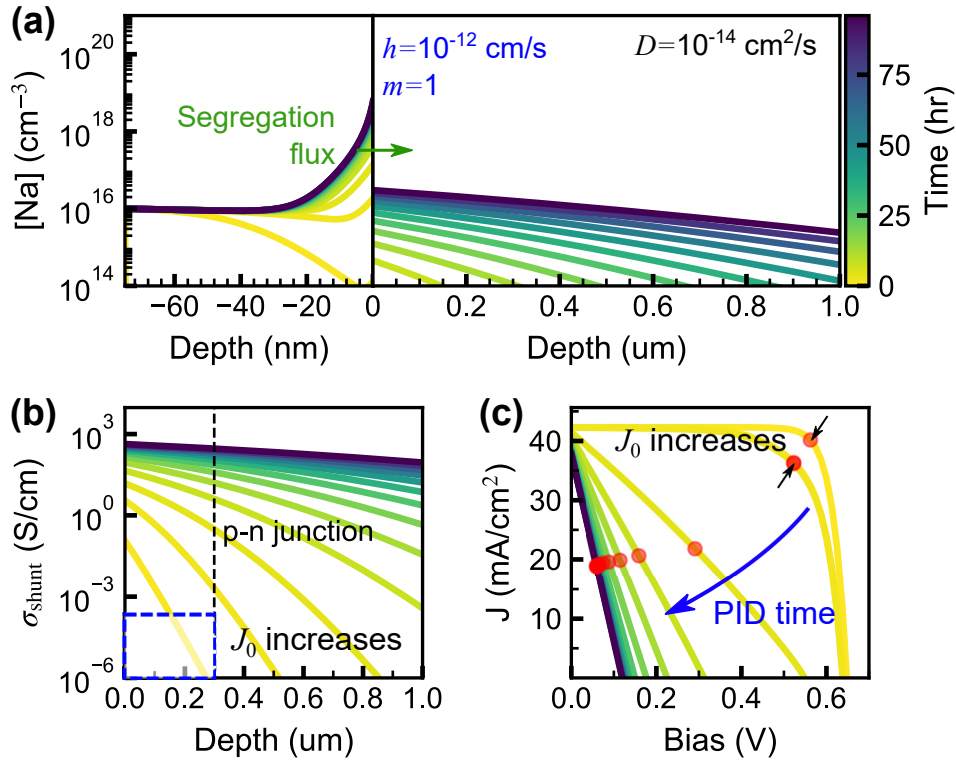


Figure 5.5: Simulated PID-s kinetics degradation for device stressed at  $85^\circ\text{C}$  under a voltage equivalent to  $0.5\text{ MV cm}^{-1}$ . (a) Shows the corresponding Na concentration profiles as a function of PID stress indicated in color scale. Accumulation at the Si interface occurs due to the effect of the segregation flux allowing only a fraction of Na to sink into the  $n^+$ -Si emitter. In Si, Na transport is entirely governed by diffusion. For the selected value of the diffusion coefficient of Na in the SF ( $10^{-14}\text{ cm}^2/\text{s}$ ), Na reaches the  $p$ - $n$  junction of the devices in a few hours. (b) The corresponding conductivity from the Na concentration in the Si emitter, estimated from spreading resistance measurements by Korol [97], (c) Simulated J-V characteristic as a function of PID time for device. For short times, before the conductivity reaches the value of  $10^{-3}\text{ S cm}^{-1}$ , the device loses efficiency by increased recombination. Indicated in red circles is the maximum power point  $P_{\text{mpp}}$  for each curve.

1051 loss of fill factor is observed. This is assumed to be due to an increase in the diode recombination  
1052 current. Once this threshold is surpassed, an unambiguous decrease in  $R_{sh}$  is seen at further times,  
1053 as expected in PID-s. Note that the  $42 \text{ mA/cm}^2$  short circuit current ( $J_{sc}$ ) is due to the fact that no  
1054 surface recombination current is considered for the contacts of the cell. This fact does not change  
1055 the analysis, since the kinetics is estimated normalizing with respect to the initial performance  
1056 metrics (e.g.  $R_{sh}$  and  $P_{mpp}$ ) of the device.

## 1057 **5.4 Bounding the value of the diffusion coefficient of Na in the**

### 1058 **SF**

1059 From the simulated J-V characteristic like the one shown in Fig. 5.5, it can be seen that a  
1060 threshold value of the conductivity at the vicinity of the  $p-n$  junction is required in order to observe  
1061 PID. This indicates that, in order to observe PID, the diffusion length of Na in Si should be around  
1062 the value of the depth of the  $p-n$  junction. It follows that this requirement can be used to set the  
1063 bounds of the diffusivity of Na in the stacking fault, by using order of magnitude analysis.

1064 Robust device simulations require full 3D implementation to avoid current collection losses  
1065 artificially introduced by reduction of current pathways in 2D simulations. This increases the com-  
1066 putational load due to the large number of elements required to adequately capture the physics  
1067 on elements of very dissimilar scales. For instance, the dimensions of the quasi-2D shunt give  
1068 elements with minimum mesh size of 5 nm, while the distance between finger electrodes is 1 mm.  
1069 To reduce computational time, when working with numerous parameter scans, we recreated the  
1070 FEM model, by means a, Machine Learning algorithm that uses the concentration profiles as pre-

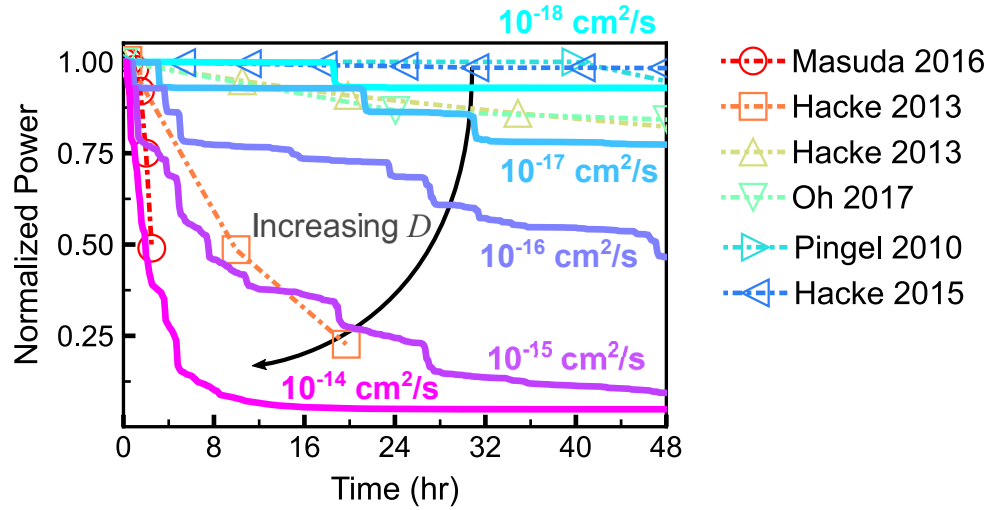


Figure 5.6: Simulated PID-s progression as a function of Na diffusivity in the stacking fault  $D_{SF}$  for a PV module stressed at 85 °C and a voltage stress equivalent to a 0.5 MV cm<sup>-1</sup> in the SiN<sub>x</sub> film. Failure time decreases with increasing  $D_{SF}$ , from being less than 10 %, within the timescale of the simulation for  $D_{SF} = 10^{-18}$  cm<sup>2</sup>/s, to less than 4 h in the case of  $D_{SF} = 10^{-14}$  cm<sup>2</sup>/s. For reference, experimental PID-s reported for similar temperatures and device architectures are plotted alongside: Masuda et al. [35] (2016), Hacke et al. [36] (2013), Oh et al. [37] (2017), Pingel et al. [5] (2010), Hacke et al. [38] (2015).

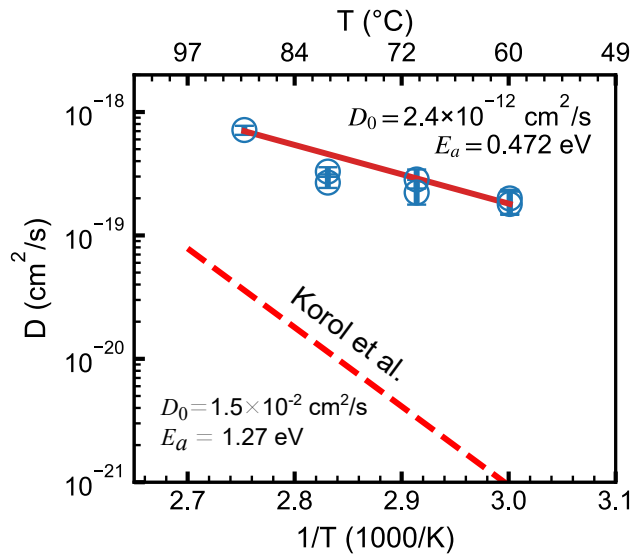


Figure 5.7: Arrhenius relationship for the diffusion coefficient of Na in Si, as estimated from fitting SIMS profiles of thermally diffused Na in n-Si wafers. For reference, the dashed red curve represents an extrapolation from from Arrhenius curves within the range of 350 to 800 °C by [97]. Adapted from Clenney et al. [109].

1071 dictors and previously computed  $P_{\text{mpp}}$  as targets. This is justified, since device simulations do not  
1072 explicitly incorporate ion transport kinetics, and the characteristic J-V is only a function of the  
1073 shunt conductivity for the geometry shown in Fig. 5.4, albeit this approach is specific to the device  
1074 geometry defined in 5.3. To implement this, we used the Scikit-learn [110] library distributed with  
1075 Python to train a Random Forest regressor with a set of 121 concentration profiles ( $\approx 70\%$ ) and their  
1076 respective  $P_{\text{mpp}}$  from a total sample size of 172 simulations. We tested the regressor against the re-  
1077 maining dataset of 52 profiles ( $\approx 30\%$ ) with their respective  $P_{\text{mpp}}$ , from which we got a coefficient  
1078 of determination [111]  $R^2 = 0.999$  for  $P_{\text{mpp}}$ .

1079 In order to bound the values of  $D_{\text{SF}}$ , we used the trained model to predict the decay in  
1080 the maximum power from the simulated concentration profiles of devices stressed according to  
1081 the parameters in Table 5.1. The diffusion coefficient of Na in the SF is varied logarithmically  
1082 from  $10^{-18}$  to  $10^{-14}$   $\text{cm}^2/\text{s}$ . Fig. 5.6 shows the simulated power output ( $P_{\text{mpp}}$ ) normalized to  
1083 the initial  $P_{\text{mpp}}$  as a function of PID time, for different values of  $D_{\text{SF}}$ . As expected, failure time  
1084 decreases with increasing  $D_{\text{SF}}$ , from a few hours for  $D_{\text{SF}} = 10^{-14}$   $\text{cm}^2/\text{s}$ , to the order of days for  
1085  $D_{\text{SF}} = 10^{-18}$   $\text{cm}^2/\text{s}$ . Overlaid in Fig. 5.6 are experimental degradation times reported for p-Si  
1086 PV modules stressed under similar conditions reported by Pingel et al. [5], Masuda et al. [35],  
1087 Hacke et al. [36], Oh et al. [37], and Hacke et al. [38]. Comparison with literature data for PID-  
1088 susceptible devices like the ones reported by Masuda et al. [35], indicates that the upper bound  
1089 for  $D_{\text{SF}}$  is of the order of  $10^{-14}$   $\text{cm}^2/\text{s}$ . It is notable that the devices which experience faster  
1090 degradation kinetics (Masuda et al. [35] and Hacke et al. [36]) are multi-crystalline Si modules,  
1091 whereas devices exhibiting smaller degradation rates (Pingel et al. [5], Oh et al. [37], and Hacke et  
1092 al. [38]) are mono-crystalline devices. Recent results from *ab-initio* calculations on the interstitial  
1093 barrier for Na diffusion in intrinsic stacking faults suggested that diffusion along the extend defects

1094 might have a larger energy penalty than bulk diffusion [112]. This result seemed to indicate that the  
1095 mechanism behind shunt decoration is dominated by bulk diffusion followed by segregation to the  
1096 intrinsic SF. Fig. 5.7 shows the Arrhenius relationship of Na diffusivity in Si as obtained by fitting  
1097 the diffusion profiles of Na measured by SIMS [109]. Based on SIMS measurements, the expected  
1098 diffusivity of Na in the bulk of Si is  $\approx 3 \times 10^{-19} \text{ cm}^2/\text{s}$  at  $85^\circ\text{C}$  [109], which contrasts with the  
1099 upper bound of  $10^{-14} \text{ cm}^2/\text{s}$  required to match the degradation kinetics seen on multicrystalline  
1100 samples [35]. Our result indicates that segregation to the SF from bulk-diffused Na is incompatible  
1101 with the timescales associated with degradation in PID-prone devices. This suggests a diffusive  
1102 mechanism through the SF as described by the DFT calculations by Ziebarth et al. [113], in which  
1103 the energetic barrier for interstitial diffusion along the SF is lowered for a filled defect, driven by  
1104 an increase in the Si–Si bond distance.

## 1105 **5.5 Finite Source Analysis**

1106 The considerable amount of Na accumulation in  $\text{SiN}_x$  near the Si interface shown in Fig.  
1107 5.5 (reaching concentrations  $> 10^{18} / \text{cm}^{-3}$ ) is in part due to the low value of  $h$  ( $10^{-12} \text{ cm/s}$ ) used  
1108 in the simulation. However, a significant contribution to the accumulation comes from the fact that  
1109 we have assumed an infinite reservoir of  $\text{Na}^+$  keeping the concentration  $C_s$  fixed at all times. A  
1110 more physical assumption is to consider the effect of a finite source by means of Eq. (4.18),

$$J_{\text{source}} = kS$$

1111 where a constant flux of  $\text{Na}^+$  at the source is given by the product of a rate of ingress  $k$  and a  
1112 surface concentration  $S$ .

Table 5.2: Parameters taken as the reference base case in PID kinetics simulations

Parameter	Value	Units	Consideration
$T$	85	°C	Accelerated testing
$E$	$10^4$	V/cm	Within the order of magnitude of the upper bound in Table 2.1.
$D_{\text{SiN}}$	$4 \times 10^{-16}$	$\text{cm}^2/\text{s}$	Ref. [62] (at 85 °C).
$D_{\text{SF}}$	$10^{-14}$	$\text{cm}^2/\text{s}$	Upper bound from Sec. 5.4.
$S$	$10^{10}$	$\text{cm}^{-2}$	Two order of magnitude less than intentionally contaminated devices in Wilson et al. [32].
$k$	$10^{-4}$	$\text{s}^{-1}$	Depletion of the source within the first 24 h.
$h$	$10^{-12}$	$\text{cm s}^{-1}$	Inefficient leakage to Si
$m$	1	–	No jump discontinuity.
$L_{\text{SiN}}$	75	nm	Ref. [34]
$L_{\text{Si}}$	1	$\mu\text{m}$	Capture the emitter thickness.

1113 The value of  $k$  can be defined based upon whether depletion of the source occurs during  
1114 the time scale of the simulation or not, whereas  $S$  can be taken from experimental estimations on  
1115 the surface concentration like the ones given by Naumann et al. [104], Wilson et al. [32] and Snow  
1116 et al. [69].

1117 As shown in Fig. 4.5, the consideration of a finite source leads to (1) less accumulation at  
1118 the  $\text{SiN}_x/\text{Si}$  interface and, (2) depletion of the source within the timescale of the simulation.

1119 To understand the effect of individual variations of the parameters  $D_{\text{SF}}$ ,  $S$ ,  $k$ ,  $E$  and  $h$  from  
1120 the base case indicated in Table 5.2, we performed PID kinetics simulations using a finite source

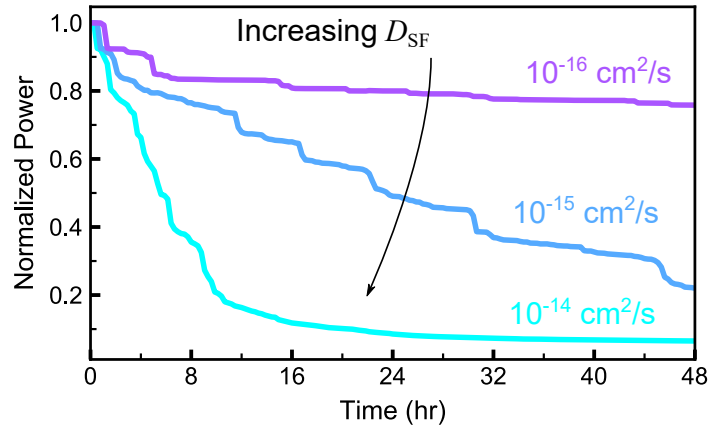


Figure 5.8: Simulated PID-kinetics as a function of  $D_{SF}$  (indicated by the color scale).

1121 boundary condition and varying one parameter at a time in logarithmic intervals.

1122 In Fig. 5.8, the simulated kinetics of PID-s is shown for different values of  $D_{SF}$ , indicated  
 1123 in color scale. As could be expected, larger values of  $D_{SF}$  result in a more rapid accumulation of Na  
 1124 at the  $p-n$  junction of the device leading to faster degradation rates. This is by far the parameter that  
 1125 induces the most significant variation in the degradation kinetics of p-Si PV modules. Depending  
 1126 on the quality of the initial wafer and the manufacturing quality control during the formation of the  
 1127 emitter, the number of SF defects can vary from device to device, leading to effective variations  
 1128 in  $D_{SF}$ . Furthermore, it is expected for multi-crystalline Si devices to have grain boundaries with  
 1129 different defect orientations, with respect to the typical  $54^\circ$  orientation of the  $\{111\}$  SF in mono-  
 1130 crystalline Si. Some of these orientations would lead to shorter SF path lengths to the emitter  
 1131 junction. This could explain difference on the effective  $D_{SF}$ , which would be averaged over the  
 1132 crystallite boundaries at the surface of the emitter.

1133 In Fig. 5.9 the simulated time series of PID is plotted for different values of the surface  
 1134 source concentration,  $S$  indicated by the color scale. It can be observed that even for a variation



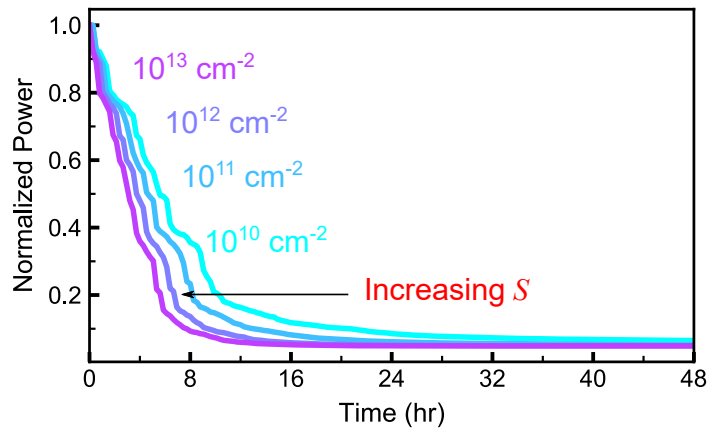


Figure 5.9: Simulated PID-kinetics as a function of the surface concentration of Na at the source  $S$  (indicated by the color scale).

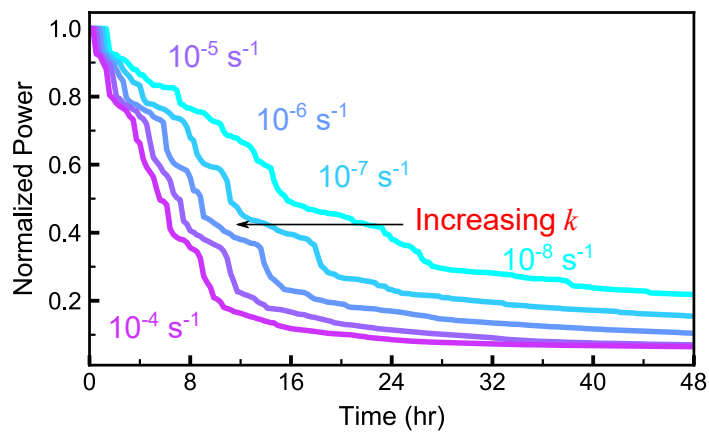


Figure 5.10: Simulated PID-kinetics as a function of the rate of ingress of Na at the source  $k$  (indicated by the color scale)

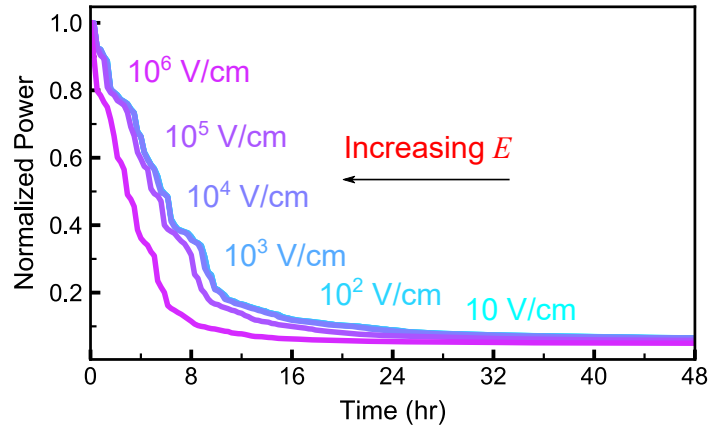


Figure 5.11: Simulated PID-kinetics as a function of the magnitude of the electric field in  $\text{SiN}_x$   $E$  (indicated by the color scale).

1135 of three orders of magnitude in  $S$ , the equivalent time required to reach a 50 % degradation of the  
 1136 initial power output is still within the range of 4 to 8 hours. Nevertheless, the degradation reaches  
 1137 almost the same level, regardless of the value of  $S$ , after  $\sim 20$ h of PID stress.

1138 In contrast, in Fig. 5.10 it is shown that the rate of ingress of Na to the  $\text{SiN}_x$  has a significant  
 1139 impact on the magnitude and the rate of degradation of the PV module. It can be seen that the  
 1140 time required to reach a 50 % degradation varies from around 16 h for  $k = 10^{-8} \text{ s}^{-1}$ , to  $\approx 8$  h for  
 1141  $k = 10^{-5} \text{ s}^{-1}$ . An increase in the flux from the source leads to a larger accumulation of Na in  
 1142 the  $\text{SiN}_x$  film, compared with lower values of  $k$ . Since the magnitude of the segregation flux is  
 1143 dependent on the difference of the Na concentration at both sides of the  $\text{SiN}_x/\text{Si}$  interface, a larger  
 1144 value of the concentration at the  $\text{SiN}_x$  increases the magnitude of the flux to the Si emitter. This  
 1145 explains why larger values of  $k$  increase the rate and the magnitude of degradation of the module.

1146 As discussed in Sec. 1.4, it is expected that the magnitude of the electric field  $E$  in  $\text{SiN}_x$   
 1147 determines the amount of drift of Na through the dielectric and hence, a reduction of  $E$  should  
 1148 translate to a decrease in the decay rate. In Fig. 5.11 the simulated PID time series of a PV modules

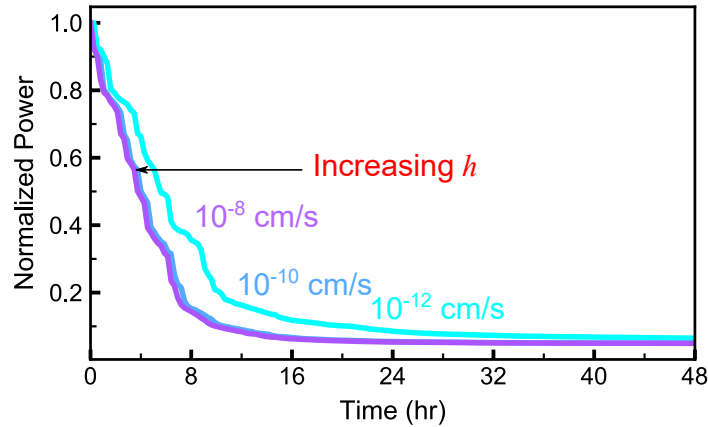


Figure 5.12: Simulated PID-kinetics as a function of the magnitude of the surface mass transfer coefficient  $h$  at the  $\text{SiN}_x/\text{Si}$  interface (indicated by the color scale).

1149 is shown, as a function of  $E$ , indicated in color scale. It can be observed that, indeed, a reduction of  
 1150 the  $E$  from  $10^6$  to  $10^5$  MV/cm leads to a  $< 4$  h delay in the time the device is expected to degrade  
 1151 50 % with respect to the initial value. Nevertheless, variations in the degradation kinetics for for  
 1152  $E < 10^4$  MV/cm are minimal. This behavior is explained by the fact that, at  $85^\circ\text{C}$ , the diffusion  
 1153 coefficient of  $\text{Na}^+$  in  $\text{SiN}_x$  is such, that transit time can occur in less than 36 h for  $E < 10^4$  MV/cm,  
 1154 as can be seen in Fig. 3.4. This indicates that it is likely that highly resistive  $\text{SiN}_x$  films preclude the  
 1155 fabrication of PID-robust devices. It might be possible that the energetic barrier for  $\text{Na}^+$  diffusion  
 1156 in less resistive  $\text{SiN}_x$  can be larger compared to high resistivity  $\text{SiN}_x$ , leading to a decrease in the  
 1157 rate of decay with respect to the indicated in Fig. 5.11.

1158 In Fig. 5.12, the variation in the simulated PID kinetics as a function of the surface mass  
 1159 transfer coefficient  $h$  at the  $\text{SiN}_x/\text{Si}$  interface is shown. It can be observed that for  $h > 10^{-10}$  cm/s,  
 1160 the difference in the time series of PID becomes negligible. The effect of  $h$  is conflated with the  
 1161 relative values of  $D$  in the  $\text{SiN}_x$  film and in the SF, as well as with the relative Na concentrations at  
 1162 both sides of the  $\text{SiN}_x/\text{Si}$  interface.

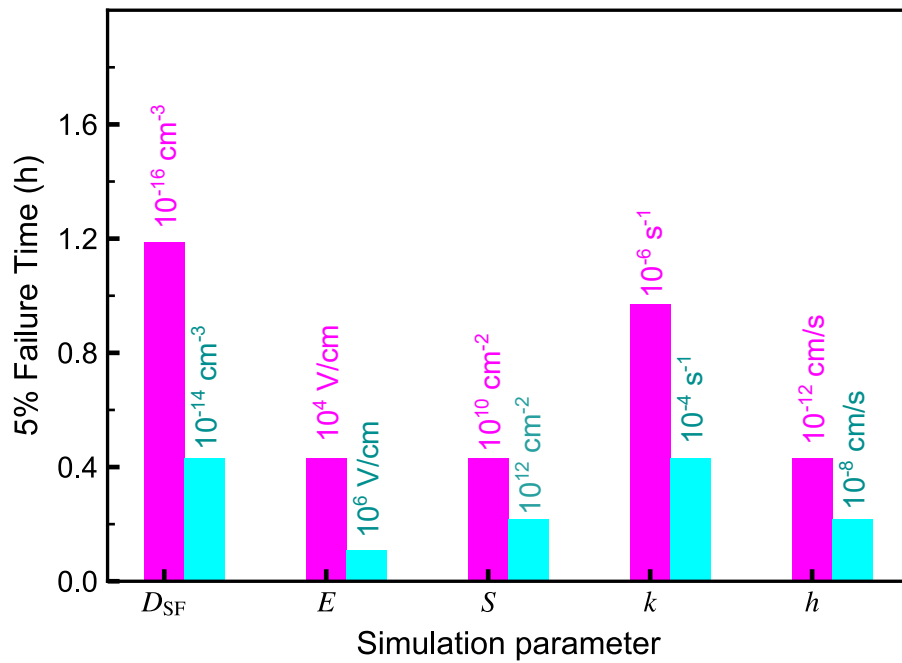


Figure 5.13: Estimated time required to reach a 5 % performance degradation in a p-Si PV module due to potential induced degradation as a function of (1) the diffusion coefficient of Na in the stacking fault  $D_{SF}$ , (2) the electric field in the  $\text{SiN}_x$   $E$ , (3) the surface concentration of ions at the source  $S$ , (4) the rate of ingress of Na from the source  $k$ , (5) the segregation surface mass transfer coefficient at the  $\text{SiN}_x/\text{Si}$  interface  $h$ . For comparison, a “high” and a “low” value of each parameter is shown with a 2 orders of magnitude variation in each case.

1163 To quantitatively compare the effect of the variation in the transport parameters, we esti-  
1164 mate the time required to incur in a 5 % decrease in the initial power output ( $\tau_5$ ) for each of the  
1165 time series shown in Figs. 5.8, 5.9, 5.10, 5.11 and 5.12. The comparison is made for a two orders  
1166 of magnitude variation of each parameter with respect to the base case shown in Tab. 5.2. The  
1167 values are plotted in Fig. 5.13. It can be seen that  $\tau_5$  decreases almost 75 % when increasing  $E$   
1168 from  $10^4$  to  $10^6$  V/cm which is the largest percentage change between the five parameters. Nev-  
1169 ertheless, it has been shown that below  $10^4$  V/cm, diffusive transport dominates leading to a  
1170 negligible change in the kinetics of PID-s, for the  $\text{SiN}_x$  composition analyzed in this work. The  
1171 second largest change occurs when varying  $D_{\text{SF}}$  from  $10^{-16}$  to  $10^{-14}$   $\text{cm}^2/\text{s}$ , which results in a  
1172  $\approx 64\%$  reduction in  $\tau_5$ . Similarly  $\tau_5$  drops  $\approx 55\%$ , when increasing  $k$  from  $10^{-6}$  to  $10^{-4}$   $\text{s}^{-1}$ .  
1173 Finally,  $\tau_5$  drops  $\approx 50\%$  when increasing  $S$  from  $10^{10}$  to  $10^{12}$   $/\text{cm}^2$ , which is just about the same  
1174 amount it decreases when varying  $h$  from  $10^{-12}$  to  $10^{-8}$   $\text{cm}/\text{s}$ . These results indicate that control  
1175 over the electric field in  $\text{SiN}_x$  is very important, particularly in film compositions that have lower  
1176 Na diffusivities. Additionally the diffusion coefficient of Na remains a significant parameter which  
1177 could be subject of improvement by using monocrystalline Si wafers and through optimization of  
1178 the process of emitter diffusion [9]. A final remark is that, the considerable change in  $\tau_5$  introduced  
1179 by the variation of  $k$  suggests that engineering the encapsulant/ $\text{SiN}_x$  interface might be a potential  
1180 workaround to increase segregation at the polymer, and hence increase the robustness of the device  
1181 against PID.

## 1182 **5.6 Summary**

1183 We developed a framework to model PID-s degradation kinetics that describes the kinetics  
1184 of shunt formation and the decay in the power output as a function of the diffusivity of Na in  
1185 the in the PV module, surface concentration and rate of ingress of Na at the EVA/SiN<sub>x</sub> interface,  
1186 segregation surface mass transfer coefficient, and the string bias. This parametrization can be  
1187 correlated with the bill of materials in the module. Extension to estimate the kinetics of  $R_{sh}$  decay  
1188 is straightforward. We identified that the shunt conductivity around the  $p-n$  junction of the device  
1189 must be above  $10^{-3} \text{ S cm}^{-1}$  in order to observe PID. Furthermore, we established an upper bound  
1190 for the diffusivity of Na in the stacking fault on the order of  $10^{-14} \text{ cm}^2/\text{s}$ , which suggests that  
1191 diffusion occurs along the SF.

1192 We established that finite source simulations could be advantageous to avoid unphysical  
1193 accumulation of Na across the interface. By simulating the effect of varying one parameter at a  
1194 time, we were able to identify that  $D_{SF}$  has the largest impact on the kinetics of PID. This indicates  
1195 that the crystallinity and number concentration of SF in the emitter can average to an effective  
1196 diffusion coefficient. It is expected then, that quality control during the emitter formation could  
1197 help decrease the susceptibility to PID. Notably, the kinetics of PID is very sensitive to the rate  
1198 of ingress of Na in the SiN<sub>x</sub>. This means that interface design between the encapsulant and SiN<sub>x</sub>  
1199 is very likely to reduce the impact of PID in Si PV modules, by increasing segregation to the  
1200 encapsulant. We also identified that, for highly resistive SiN<sub>x</sub>, the reduction of  $E$  has a limiting  
1201 effect on PID, due to the large value of the diffusivity of Na associated to the Si-N composition we  
1202 studied.

1203 Overall, this framework has enabled us to identify the rate limiting processes in PID. Fur-  
1204 ther fitting of the degradation time series to accelerated PID experiments will allow us to deter-  
1205 mine more accurate values on the parameter space. Our model provides a description of the PID-s  
1206 physics which renders not only the time series of the power decay, but also the kinetics of Na  
1207 transport and J-V characteristic of stressed devices. This will enable reference with multiple char-  
1208 acterization techniques (e.g. BTS measurements, SIMS characterization, I-V characterization), to  
1209 validate the sensitivity to the different components in the module and. Upon parameter refinement,  
1210 this framework will be of use in the design of PID-robust PV modules.

1211 Chapter 5, in part, is being prepared for submission in *IEEE Journal of Photovoltaics* un-  
1212 der the title “Finite Element Simulation of Potential Induced Degradation Kinetics in p-Si Solar  
1213 Modules” by E. Martinez-Loran, G. von Gastrow, J. Clenney, R. Meier, P. Bandaru, Mariana I.  
1214 Bertoni, D.P. Fenning. The dissertation/thesis author was the primary investigator and author of  
1215 this paper.

## 1216 **6 | Summary and Suggestions for Future**

### 1217 **Work**

1218 A framework for the characterization and modeling of potential-induced degradation has  
1219 been developed for p-Si photovoltaic modules. The proposed methodology includes quantification  
1220 methods to determine the transport kinetics of Na in the passivation layers of Si solar cells using  
1221 measurement techniques that enable characterization at device operating temperatures. The advan-  
1222 tage of this measurement is that it can provide a more accurate parametrization of ion kinetics with  
1223 respect to the extrapolation of the diffusivities from measurements at higher temperature ranges,  
1224 which can incur orders of magnitude variations in the estimated diffusion coefficient.

1225 Transport in  $\text{SiN}_x$  films with index of refraction 1.89 (at 600 nm) reveals that anti-reflective  
1226 coatings with this Si-N composition, are not an effective barrier against Na diffusion in silicon  
1227 photovoltaic modules.

1228 A numerical solution to the transport of charged species in dielectrics was developed which  
1229 can describe the ingress of Na contamination in silicon photovoltaic modules. The advantage of the  
1230 numerical solution is that: (1) it can solve Poisson and Nernst-Planck's equations self consistently  
1231 to completely describe the kinetics of Na transport for contamination levels consistent with PID,



1232 (2) extension of the solution to stacks of materials can be implemented through first order kinetics  
1233 segregation fluxes [96], (3) It allows to reference the numerical results with electrical measure-  
1234 ments and other quantitative experimental techniques to characterize the concentration of Na ions  
1235 (e.g. secondary-ion mass spectroscopy).

1236 A complete framework to describe the kinetics of potential-induced degradation was pro-  
1237 posed which uses transport simulations to estimate the resistivity profile of a metallic shunt in the  
1238 emitter of a photovoltaic module. This model has allowed us to set an upper bound of  $10^{-14}$  cm<sup>2</sup>s  
1239 (at 85 °C) on the diffusivity of Na in the Si emitter. This result indicates that bulk diffusion,  
1240 which occurs with  $D \approx 3 \times 10^{-19}$  cm<sup>2</sup>/s at the simulated temperature [109], is not the predomi-  
1241 nant mechanism in PID-prone devices. Instead, transport is dominated by diffusion through the SF  
1242 as suggested by *ab-initio* studies by Ziebarth et al. [113].

## 1243 **6.1 Applications of PID kinetics quantification and modeling**

1244 The methodology developed to quantify the kinetics of Na transport in SiN<sub>x</sub> is of interest to  
1245 other device architectures and dielectric compositions. Determination of transport kinetics is ap-  
1246 plicable to other semiconductor architectures like perovskite PV devices, which are very attractive  
1247 for solar energy harvesting [114].

1248 The general goal of this modeling framework is to be able to parametrize PID-s in terms  
1249 of the materials used in the photovoltaic module and the operating conditions. This model will  
1250 provide guidance in the design of ion blocking layers that can reduce the ingress of Na to the  
1251 emitter of the module by means of (1) reduction of the drift and, (2) interface design to increase  
1252 segregation on the outermost layers of materials in the device.

## 1253 **6.2 Future work**

### 1254 **6.2.1 Quantification of composition-dependent transport in SiN<sub>x</sub>**

1255 Based on the voltage divider model, modification of the Si to N composition in SiN<sub>x</sub> films  
1256 impacts PID kinetics by changing the electrical resistivity of the film and hence, the magnitude of  
1257 the electric field driving transport. Nevertheless, transport of Na can also be influenced by local  
1258 electrostatic interactions determined by the coordination environment of (N–Si–N)<sup>-</sup> in the film  
1259 [115]. To quantify this effect, the trapping corrected BTS method can be used to study sodium  
1260 transport in SiN<sub>x</sub> films as a function of Si to N composition. It would be expected that SiN<sub>x</sub> of  
1261 higher Si content [33] would experience a higher activation energy for diffusion.

### 1262 **6.2.2 Incorporation of environmental factors**

1263 To adequately reproduce accelerated testing protocols, incorporation of temperature- and  
1264 humidity-activated resistivity [3] values is required. This could help elucidate whether transport  
1265 in the polymer encapsulants can be enhanced by an electrochemical reaction at elevated water  
1266 contents, in addition to the electrostatic considerations of this model.

1267 PID kinetics simulations will be of use to better understand the applicability of accelerated  
1268 testing protocols. Accelerated testing assumes long term damage can be characterized by contin-  
1269 uous application of temperature, voltage and humidity stress to the PV modules. Under regular  
1270 operation, such extreme conditions do not occur. Diurnal variations can be implemented on our  
1271 transport modeling framework to account for realistic ingress of Na under conditions closer to field  
1272 operation.

### 1273 **6.2.3 Three-dimensional considerations**

1274           The present model assumes that Na transport is predominantly occurring in the direction  
1275 normal to the material interfaces which simplifies some of the derivations and saves computing  
1276 resources. Nevertheless, three-dimensional extension can be easily implemented, and can provide  
1277 further information on the effect of in-plane diffusion, especially in the context of interfacial oxide  
1278 layers that might play a significant role in reducing the ingress of Na contamination to the Si  
1279 emitter.

### 1280 **6.2.4 PID-recovery**

1281           The process of PID recovery has been reported experimentally in p-Si PV modules by  
1282 means of the application of a thermal stress with and without the reversal of the electric field in the  
1283 PV module [5, 11, 14, 15, 28, 35, 103, 116, 117]. While full PID recovery is usually not observed,  
1284 the process remains poorly understood from a physical perspective.

1285           The present model considers that transport of Na impurities in the Si emitter of PV de-  
1286 vices is dominated by 1D diffusion along the direction normal to the  $\text{SiN}_x/\text{Si}$  interface. Under  
1287 these conditions, the typical negative concentration gradients inside the silicon emitter preclude  
1288 out-diffusion as proposed by Lausch et al. [11]. In the context of this model, we hypothesize  
1289 that recovery could be possible for a specific combination of source concentrations and recovery  
1290 electric fields. A full combinatorial analysis on the parameter space could shed light on such con-  
1291 ditions. An alternative answer might arise from the three-dimensional consideration of transport  
1292 kinetics, where out-diffusion to the bulk of Si might explain thermal PID recovery. In this sense,  
1293 Density Functional Theory studies have shown that, once Na occupies interstitial positions in the

1294 SF, there is a large barrier to diffuse back into the bulk [112], which explains the irreversibility of  
1295 this degradation mode.

## 1296 **6.2.5 Incorporation of SiN<sub>x</sub> trap kinetics**

1297 In our trapping corrected BTS method we considered the effect of bulk traps in SiN<sub>x</sub> cap-  
1298 turing, driven by the applied voltage stress. The rate at which traps are filled is dependent on the  
1299 magnitude of the applied field, and the energy position of the traps with respect to the conduction  
1300 band edge of SiN<sub>x</sub> as prescribed by Frenkel-Poole emission current [63]:

$$J = C_1 E \exp \left\{ -\frac{q}{k_B T} \left[ \phi_1 - \left( \frac{qE}{\pi \epsilon_0 \epsilon_d} \right)^{1/2} \right] \right\}, \quad (6.1)$$

1301 where  $E$  is the electric field,  $\phi_1$  is the barrier height of the trap with respect to the conduction band  
1302 edge,  $\epsilon_d$  is the dynamic dielectric constant of SiN<sub>x</sub> and the proportionality constant  $C_1$  is a function  
1303 of the density of the trapping centers.

1304 During operating conditions, the voltage differential in the SiN<sub>x</sub> anti-reflective coating,  
1305 albeit low, might be enough to fill a finite concentration of bulk traps with electrons leading to  
1306 positive image charges in the semiconductor. Based upon the 1 V saturation flatband voltage shift  
1307 (see Fig. 3.2) from our BTS measurements at 1 MV cm<sup>-1</sup> and Eq. (6.1), we assumed that the  
1308 concentration of occupied traps at  $E = 10$  kV cm<sup>-1</sup> typically used for our simulations, will be at  
1309 least two orders of magnitude less. Hence, we neglected the concentration of trapping centers in  
1310 the computation of image charges in PID degradation kinetics. Nevertheless, this aspect could  
1311 be of interest for SiN<sub>x</sub> compositions with larger concentration of traps. To characterize this, we  
1312 propose to fit  $\Delta V_{FB}$  in SiN<sub>x</sub> MIS capacitors without intentional Na contamination to the flatband  
1313 voltage obtained by integration of Frenkel-Poole current in Eq. (6.1), to extract  $C_1$  and  $\phi_1$ . This

1314 will allow us to determine the extent of positive image charges in Na transport simulations, which  
1315 will modify the electric potential at the SiN<sub>x</sub>/Si interface and consequently Na accumulation and  
1316 ingress to the Si emitter.

# 1317 **A | Derivation for the characteristic time for** 1318 **drift-diffusion**

1319 We defined the characteristic time for drift-diffusion as the solution to the equation

$$L = 2\sqrt{D\tau_c} + \mu E \tau_c$$

1320  $\tau_c$  is then given by the solution to the quadratic equation

$$\mu^2 E^2 \tau_c^2 - 2(\mu E L + 2D)\tau_c + L^2 = 0, \quad (\text{A.1})$$

1321 which is given by

$$\tau_c = \frac{L}{\mu E} + \frac{2D}{\mu^2 E^2} \pm 2 \frac{\sqrt{\mu E L D + D^2}}{\mu^2 E^2}. \quad (\text{A.2})$$

1322 Rearranging the terms within the square root term:

$$\tau_c = \frac{L}{\mu E} + \frac{2D}{\mu^2 E^2} \pm \frac{2D}{\mu^2 E^2} \left(1 + \frac{\mu E L}{D}\right)^{1/2}. \quad (\text{A.3})$$

1323 Equation (A.3) can be further simplified to

$$\tau_c = \frac{L}{\mu E} + \frac{2D}{\mu^2 E^2} \left[1 \pm \left(1 + \frac{\mu E L}{D}\right)^{1/2}\right]. \quad (\text{A.4})$$

1324 Writing the mobility term explicitly ( $\mu = qD/k_B T$ ):

$$\tau_c = \frac{L}{D} \left( \frac{k_B T}{qE} \right) + \frac{2}{D} \left( \frac{k_B T}{qE} \right)^2 \left\{ 1 \pm \left[ 1 + \left( \frac{qE}{k_B T} \right) L \right]^{1/2} \right\}. \quad (\text{A.5})$$

1325 Equation (A.5) can also be expressed in term of the thermal voltage  $v_{\text{th}} \equiv k_B T/q$ :

$$\tau_c = \frac{L}{D} \left( \frac{v_{\text{th}}}{E} \right) + \frac{2}{D} \left( \frac{v_{\text{th}}}{E} \right)^2 \left\{ 1 \pm \left[ 1 + \left( \frac{E}{v_{\text{th}}} \right) L \right]^{1/2} \right\}. \quad (\text{A.6})$$

1326 The solution to the equation (3.8) assumes that the drift flux  $\mu EC$  is in fact considerable  
1327 with respect to the diffusive flux  $-D\nabla^2 C$ . Equation (A.6) depends on the inverse of  $E$ . This causes  
1328  $\tau_c \rightarrow \infty$  as  $E \rightarrow 0$ , which is not physical because the diffusive term dominates as  $E \rightarrow 0$ . Then the  
1329 solution needs to be given by the lowest bound of the time between (A.6) and the characteristic  
1330 diffusion time  $L/D$ .

# 1331 **B | Derivation of the spatial part of Nernst-**

## 1332 **Planck equation in the weak form**

1333 Behind scenes, the weak formulation of a PDE works by transforming the differential op-  
1334 erator to a variational problem [81, 82, 118]. Discretization prescribes formulating the problem  
1335  $L(x) = f$  in matrix form. The solution  $u$  in the vector space  $V$  is then given by:

$$Au = f. \tag{B.1}$$

1336 The function space  $V$  has the minimum requirements of being equipped with a norm  $\|\cdot\|$  to compute  
1337 distances within vectors, and we require that every Cauchy sequence in  $V$  converges to an element  
1338 in the same space (Banach space) [119]. The variational formulation states that this problem is  
1339 equivalent to finding  $u$  such that

$$[Au](v) = f(v) \tag{B.2}$$

1340 for all  $v$  in the vector space  $V$ . In the FEM,  $u$  are called *trial functions* and  $v$  are known as *test*  
1341 *functions*. An additional requirement is placed on  $V$  so that for all test functions  $v^2$  and  $\|v^2\|$  have  
1342 finite integrals over the volume (Sobolev space) [81].

1343 The finite-element-method approach to integrate the spatial part of the coupled Poisson-  
1344 Nernst-Planck system with the finite element method is to write it in the variational formulation.



1345

The right hand side of Eq. (2.1) and Poisson's equation can be written in terms of trial

1346 functions

$$L_{\text{NP}}[u_c] \equiv \nabla \cdot (D\nabla u_c) + \mu u_c \nabla \cdot \nabla u_p + \mu \nabla u_c \cdot \nabla \phi(u_p, t), \quad (\text{B.3a})$$

$$L_{\text{P}}[u_p] \equiv \nabla \cdot \nabla u_p + \frac{q}{\epsilon_{\text{ins}}} u_c \quad (\text{B.3b})$$

1347

where  $u_c$ ,  $u_p$  are trial functions for the concentration and electric potential respectively,  $L_{\text{NP}}$  and

1348

$L_{\text{NP}}$  are the differential operators for the spatial part of (2.1) and for Poisson's equation, respec-

1349 tively

1350

Multiplying  $L$  by the test function  $v$  and integrating over the volume  $\Omega$ :

$$A_{\text{NP}}(u, v) \equiv D \int_{\Omega} \nabla \cdot (\nabla u_c) v_c d\Omega + \mu \int_{\Omega} u_c \nabla \cdot \nabla u_p v_c d\Omega + \mu \int_{\Omega} \nabla u_c \cdot \nabla u_p v_c d\Omega \quad (\text{B.4a})$$

$$A_{\text{P}}(u, v) \equiv \int_{\Omega} \nabla \cdot \nabla u_p v_p d\Omega + \frac{q}{\epsilon_{\text{ins}}} \int_{\Omega} u_c v_p d\Omega \quad (\text{B.4b})$$

1351

1352

The order of the derivatives of Laplacian terms in (B.4) can be reduced by integrating by

1353

parts and using Gauss theorem (See Appendix D):

$$A_{\text{NP}}(u, v) \equiv -D \int_{\Omega} \nabla u_c \cdot \nabla v_c d\Omega + D \int_{\partial\Omega} (\nabla u_c \cdot \hat{\mathbf{n}}) v_c ds - \mu \int_{\Omega} \nabla u_p \cdot \nabla (u_c v_c) \Omega \\ + \mu \int_{\partial\Omega} (\nabla u_p \cdot \hat{\mathbf{n}}) u_c v_c ds + \mu \int_{\Omega} \nabla u_c \cdot \nabla u_p v_c d\Omega, \quad (\text{B.5})$$

1354

for  $A_{\text{NP}}(u, v)$  and

$$A_{\text{P}}(u, v) \equiv - \int_{\Omega} (\nabla u_p \cdot \nabla v_p) d\Omega + \int_{\partial\Omega} (\nabla u_p \cdot \hat{\mathbf{n}}) v_p ds + \frac{q}{\epsilon_{\text{ins}}} \int_{\Omega} u_c v_p d\Omega \quad (\text{B.6})$$

1355

for  $A_{\text{P}}(u, v)$ .

1356

Expanding  $\nabla(u_c v_c) = u_c \nabla v_c + v_c \nabla u_c$  in (B.5):

$$A_{\text{NP}}(u, v) \equiv -D \int_{\Omega} \nabla u_c \cdot \nabla v_c d\Omega + D \int_{\partial\Omega} (\nabla u_c \cdot \hat{\mathbf{n}}) v_c ds - \mu \int_{\Omega} \nabla u_p \cdot u_c \nabla v_c \Omega$$

$$- \mu \int_{\Omega} \nabla u_p \cdot v_c \nabla u_c \Omega + \mu \int_{\partial\Omega} (\nabla u_p \cdot \hat{\mathbf{n}}) u_c v_c ds + \mu \int_{\Omega} \nabla u_c \cdot \nabla u_p v_c d\Omega,$$

1357

The bilinear forms can now be written as

$$A_{\text{NP}} = -D \int_{\Omega} \nabla u_c \cdot \nabla v_c d\Omega + D \int_{\partial\Omega} (\nabla u_c \cdot \hat{\mathbf{n}}) v_c ds - \mu \int_{\Omega} u_c \nabla u_p \cdot \nabla v_c \Omega + \mu \int_{\partial\Omega} (\nabla u_p \cdot \hat{\mathbf{n}}) u_c v_c ds$$

(B.7a)

$$A_{\text{P}} = - \int_{\Omega} (\nabla u_p \cdot \nabla v_p) d\Omega + \int_{\partial\Omega} (\nabla u_p \cdot \hat{\mathbf{n}}) v_p ds + \frac{q}{\epsilon_{\text{ins}}} \int_{\Omega} u_c v_p d\Omega$$

(B.7b)

1358

# 1359 C | Derivation of the TR-BDF2 time-stepping 1360 in the weak form

1361 We will use the results from Appendix B to derive the time stepping system in the varia-  
1362 tional form. We start by defining Equations (4.6) and (4.7) in terms of the trial functions for the  
1363 concentration  $u_c$  and the electric potential  $u_p$ , followed by multiplication by the test functions  $v_c$   
1364 and  $v_p$  and subsequently, by integration within the simulation volume  $\Omega$  [82].

1365 For the TR step we have

$$\int_{\Omega} u_c^{n+\gamma} v_c d\Omega - \gamma \frac{\Delta t_n}{2} \int_{\Omega} \langle F^{n+\gamma}(u_c, u_p), (v_c, v_p) \rangle d\Omega = \int_{\Omega} u_c^n v_c d\Omega + \gamma \frac{\Delta t_n}{2} \int_{\Omega} \langle F^n(u_c, u_p), (v_c, v_p) \rangle d\Omega, \quad (\text{C.1})$$

1366 and for the BDF2 step

$$\int_{\Omega} u_c^{n+1} v_c d\Omega - \frac{1-\gamma}{2-\gamma} \Delta t_n \int_{\Omega} \langle F^{n+1}(u_c, v_c), (v_c, v_p) \rangle d\Omega = \frac{1}{\gamma(2-\gamma)} \int_{\Omega} u_c^{n+\gamma} v_c d\Omega - \frac{(1-\gamma)^2}{\gamma(2-\gamma)} \int_{\Omega} u_c^n v_c d\Omega, \quad (\text{C.2})$$

1367 where  $F$  refers to the spatial part of the problem  $\partial C / \partial t = F(C, \phi, x)$ , and we have defined the inner  
1368 product

$$\langle F(u_c, u_p), (v_c, v_p) \rangle \equiv \underbrace{F_c(u_c, u_p)}_{\text{Nernst-Planck}} v_c + \underbrace{F_c(u_c, u_p)}_{\text{Poisson}} v_p \quad (\text{C.3})$$

1369 Equations (C.1) and (C.2) can be expressed in terms of the linear forms (B.7).

1370 For the TR step

$$\int_{\Omega} u_c^{n+\gamma} v_c d\Omega - \gamma \frac{\Delta t_n}{2} (A_{\text{NP}}^{n+\gamma} + A_{\text{P}}^{n+\gamma}) = \int_{\Omega} u_c^n v_c d\Omega + \gamma \frac{\Delta t_n}{2} (A_{\text{NP}}^n + A_{\text{P}}^n),$$

1371 and for the BDF2 step

$$\int_{\Omega} u_c^{n+1} v_c d\Omega - \frac{1-\gamma}{2-\gamma} \Delta t_n (A_{\text{NP}}^{n+1} + A_{\text{P}}^{n+1}) = \frac{1}{\gamma(2-\gamma)} \int_{\Omega} u_c^{n+\gamma} v_c d\Omega - \frac{(1-\gamma)^2}{\gamma(2-\gamma)} \int_{\Omega} u_c^n v_c d\Omega.$$

1372

## 1373 **D | Writing the weak form of Laplacian terms**

1374 Re-formatting the weak form for Laplacian terms  $\nabla \cdot \nabla u$  is quite useful to reduce the order  
1375 of the PDE and, to introduce Neumann boundary conditions. The starting equation for the weak  
1376 form is

$$\int_{\Omega} \nabla \cdot (\nabla u) v d\Omega \quad (\text{D.1})$$

1377 We use the result from taking the divergence of  $v\nabla u$ :

$$\nabla \cdot (v\nabla u) = v\nabla \cdot \nabla u + \nabla v \cdot \nabla u.$$

1378 to integrate by parts Eq. (D.1)

$$\int_{\Omega} \nabla \cdot (\nabla u) v d\Omega = \int_{\Omega} \nabla \cdot (v\nabla u) d\Omega - \int_{\Omega} (\nabla v \cdot \nabla u) d\Omega$$

1379 Applying the Gauss theorem to the first term on the RHS:

$$\int_{\Omega} \nabla \cdot (v\nabla u) d\Omega = \int_{\partial\Omega_D \cup \partial\Omega_N} (\nabla u \cdot \hat{\mathbf{n}}) v ds, \quad (\text{D.2})$$

1380 where  $\partial\Omega_D \cup \partial\Omega_N$  is the union of the Dirichlet and Neumann surfaces. Typically,  $v$  is defined such  
1381 that the integral over the Dirichlet part of the surface vanishes. Then

$$\int_{\Omega} \nabla \cdot (\nabla u) v d\Omega = - \int_{\Omega} (\nabla u \cdot \nabla v) d\Omega + \int_{\partial\Omega_N} (\nabla u \cdot \hat{\mathbf{n}}) v ds. \quad (\text{D.3})$$

1382 The Neumann BC  $\mathbf{g} = \nabla u$  can then be imposed on the second term on the RHS of (D.3).

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